

Chemical Reactions

Question 1: Write down the balanced chemical equations of the following:

1.1 Copper + oxygen \rightarrow copper (II) oxide

1.2 Magnesium + water \rightarrow magnesium hydroxide and hydrogen gas

1.3 Hydrochloric acid + sodium hydroxide \rightarrow sodium chloride + water

[9]

Question 2: Balance the following equations:

2.1 _____ PbF_2 + _____ $\text{H}_2\text{SO}_4 \rightarrow$ _____ PbSO_4 + _____

2.2 _____ CH_4 + _____ $\text{Br}_2 \rightarrow$ _____ CBr_4 + _____ H_2

2.3 _____ SiH_4 + _____ $\text{LiOH} \rightarrow$ _____ Si(OH)_4 + _____

2.4 _____ FeF_3 + _____ $\text{MgSO}_3 \rightarrow$ _____ Mg + _____

2.5 _____ K_3N + _____ $\text{MgBr}_2 \rightarrow$ _____ Mg_3N_2 + _____ KBr

2.6 _____ NaOH + _____ $\text{Pb(NO}_3)_2 \rightarrow$ _____ Pb(OH)_2 + _____

2.7 _____ AlF_3 + _____ $\text{Na}_2\text{CO}_3 \rightarrow$ _____ NaF + _____

2.8 _____ Al + _____ $\text{O}_2 \rightarrow$ _____

2.9 _____ BaCl_2 + _____ $\text{Na}_2\text{SO}_4 \rightarrow$ _____ BaSO_4 + _____ NaCl

2.10 _____ GaBr_3 + _____ $\text{Na}_2\text{SO}_4 \rightarrow$ _____ NaBr + _____ $\text{Ga}_2(\text{SO}_4)_3$

2.11 _____ Al(OH)_3 + _____ $\text{H}_2\text{SO}_4 \rightarrow$ _____ $\text{Al}_2(\text{SO}_4)_3$ + _____ H_2O

2.12 _____ $\text{P}_2\text{S}_2 \rightarrow$ _____ + _____

2.13 _____ $\text{Al}_2(\text{SO}_4)_3$ + _____ $\text{Ca(OH)}_2 \rightarrow$ _____ Al(OH)_3 + _____ CaSO_4

2.14 _____ C_4H_{10} + _____ $\text{O}_2 \rightarrow$ _____ CO_2 + _____ H_2O

2.15 _____ Al + _____ $\text{AuCl}_3 \rightarrow$ _____ AlCl_3 + _____ Au

2.16 _____ N_2 + _____ $\text{O}_2 \rightarrow$ _____ NO_2

2.17 _____ K_3P + _____ $\text{Ag}_3\text{PO}_4 \rightarrow$ _____ KNO_3 + _____

2.18 _____ P_4 + _____ $\text{O}_2 \rightarrow$ _____ P_2O_5

[2 x 18 = 36]

Grade 9 Natural Sciences Chemical Reaction

Raffaela Di Napoli



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