

CHEMISTRY 101 Practice Exam 1 DR. PECK

- Directions:
- (1) Choose the best answer for each multiple choice question (numbers 1 - 15). Transfer your multiple choice answers onto the scantron.
 - (2) Do not separate any of the pages. When finished, give the proctor at the front of the room both the exam and the completed scantron. Cover your exam and your work whenever possible.

NAME _____

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(NOTE: This practice exam does not contain questions based upon topics from Chapter 5. The actual exam may or may not depending upon our ability to cover Chapter 5 in lecture before the date of the first exam.)

1. Which one of the following is a weak acid?
 (a) HBr (b) HClO₄ (c) HNO₃ (d) HCl (e) HClO₂

2. What is the oxidizing agent in the following reaction?

$$6\text{KOH(aq)} + 3\text{Cl}_2\text{(g)} \rightarrow \text{KClO}_3\text{(aq)} + 5\text{KCl(aq)} + 3\text{H}_2\text{O(l)}$$
 (a) KOH (b) KCl (c) KClO₃ (d) Cl₂ (e) H₂O

3. Balance the following equation with the **smallest whole number coefficients**. What is the coefficient for HCl in the balanced equation?

$$\text{SnS}_2 + \text{HCl} \rightarrow \text{H}_2\text{SnCl}_6 + \text{H}_2\text{S}$$
 (a) 4 (b) 6 (c) 3 (d) 12 (e) 2

4. Classify the following reaction by giving **all** of these reaction type(s) that apply.

I. redox	II. combination	III. decomposition
IV. displacement	V. metathesis	

$$2\text{PbO(s)} + \text{O}_2\text{(g)} \rightarrow 2\text{PbO}_2\text{(s)}$$
 (a) I and V (b) only II (c) I and II (d) only III (e) only V

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Table of Contents Example Of Chem101 Exam Solutions

1. Understanding the eBook Example Of Chem101 Exam Solutions
 - The Rise of Digital Reading Example Of Chem101 Exam Solutions
 - Advantages of eBooks Over Traditional Books
2. Identifying Example Of Chem101 Exam Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Example Of Chem101 Exam Solutions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Example Of Chem101 Exam Solutions
 - Personalized Recommendations
 - Example Of Chem101 Exam Solutions User Reviews and Ratings
 - Example Of Chem101 Exam Solutions and Bestseller Lists

5. Accessing Example Of Chem101 Exam Solutions Free and Paid eBooks
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 - Example Of Chem101 Exam Solutions eBook Subscription Services
 - Example Of Chem101 Exam Solutions Budget-Friendly Options
6. Navigating Example Of Chem101 Exam Solutions eBook Formats
 - ePub, PDF, MOBI, and More
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 - Example Of Chem101 Exam Solutions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Example Of Chem101 Exam Solutions
 - Highlighting and Note-Taking Example Of Chem101 Exam Solutions
 - Interactive Elements Example Of Chem101 Exam Solutions
8. Staying Engaged with Example Of Chem101 Exam Solutions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Example Of Chem101 Exam Solutions
9. Balancing eBooks and Physical Books Example Of Chem101 Exam Solutions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Example Of Chem101 Exam Solutions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Example Of Chem101 Exam Solutions
 - Setting Reading Goals Example Of Chem101 Exam Solutions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Example Of Chem101 Exam Solutions
 - Fact-Checking eBook Content of Example Of Chem101 Exam Solutions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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