

Unit 1 • Observations, Models, & Experiments

SCIENTIFIC NOTATION

Change the following to Scientific Notation (maintain the number of significant figures):

1. $5.280 = \underline{5.280}$

11. $2,560 = \underline{2.56 \times 10^3}$

2. $2,000 = \underline{2 \times 10^3}$

12. $.0009 = \underline{9 \times 10^{-4}}$

3. $15 = \underline{1.5 \times 10^1}$

13. $8,900,000 = \underline{8.9 \times 10^6}$

4. $6,589,000 = \underline{6.589 \times 10^6}$

14. $.0920 = \underline{9.20 \times 10^{-2}}$

5. $70,400,000,000 = \underline{7.04 \times 10^{10}}$

15. $6,300 = \underline{6.3 \times 10^3}$

6. $.00263 = \underline{2.63 \times 10^{-3}}$

16. $.90 = \underline{9.0 \times 10^{-1}}$

7. $.00589 = \underline{5.89 \times 10^{-3}}$

17. $250 = \underline{2.5 \times 10^2}$

8. $.006 = \underline{6 \times 10^{-3}}$

18. $.006087 = \underline{6.087 \times 10^{-3}}$

9. $.400 = \underline{4.00 \times 10^{-1}}$

19. $500,000 = \underline{5 \times 10^5}$

10. $.08060 = \underline{8.060 \times 10^{-2}}$

20. $.0000000105 = \underline{1.05 \times 10^{-8}}$

Make the following Metric System conversions using "unit analysis" (you may use scientific notation):

1. $100 \text{ mg} \times \frac{1 \text{ g}}{1000 \text{ mg}} = \underline{.1} \text{ g}$

2. $20 \text{ cm} \times \frac{1 \text{ m}}{100 \text{ cm}} = \underline{.2} \text{ m}$

3. $50 \text{ L} \times \frac{1 \text{ kL}}{1000 \text{ L}} = \underline{.05} \text{ kL}$

4. $22 \text{ g} \times \frac{100 \text{ cg}}{1 \text{ g}} = \underline{2200} \text{ cg}$

5. $825 \text{ cm} \times \frac{1 \text{ m}}{100 \text{ cm}} \times \frac{1 \text{ km}}{1000 \text{ m}} = \underline{8.25 \times 10^{-3}} \text{ km}$

6. $2,350 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} = \underline{2.35 \times 10^6} \text{ g}$

7. $19 \text{ mL} \times \frac{1 \text{ L}}{1000 \text{ mL}} \times \frac{100 \text{ cL}}{1 \text{ L}} = \underline{1.9} \text{ cL}$

8. $52 \text{ km} \times \frac{1000 \text{ m}}{1 \text{ km}} = \underline{52000} \text{ m}$

9. $36 \text{ m} \times \frac{100 \text{ cm}}{1 \text{ m}} = \underline{3600} \text{ cm}$

10. $18 \text{ cm} \times \frac{1 \text{ m}}{100 \text{ cm}} \times \frac{1000 \text{ mm}}{1 \text{ m}} = \underline{180} \text{ mm}$

11. $6 \text{ g} \times \frac{1000 \text{ mg}}{1 \text{ g}} = \underline{6000} \text{ mg}$

12. $4,259 \text{ mg} \times \frac{1 \text{ g}}{1000 \text{ mg}} = \underline{4.259} \text{ g}$

Estimating Length Using Scientific Notation Answer Key

L Darling-Hammond



Estimating Length Using Scientific Notation Answer Key:

Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Estimating Length Using Scientific Notation Answer Key**

In a global inundated with displays and the cacophony of quick interaction, the profound power and mental resonance of verbal art frequently diminish in to obscurity, eclipsed by the constant onslaught of noise and distractions. However, situated within the musical pages of **Estimating Length Using Scientific Notation Answer Key**, a fascinating function of fictional splendor that pulses with organic feelings, lies an unforgettable trip waiting to be embarked upon. Composed by way of a virtuoso wordsmith, this exciting opus manuals readers on a psychological odyssey, lightly revealing the latent potential and profound affect embedded within the delicate web of language. Within the heart-wrenching expanse with this evocative evaluation, we shall embark upon an introspective exploration of the book is central themes, dissect its interesting writing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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