

## Naming Ionic Compounds

Name KEY Per \_\_\_\_\_

- |  |                            |
|--|----------------------------|
| 1. NaCl  | <u>sodium chloride</u>     |
| 2. Na <sub>2</sub> S                               | <u>sodium sulfide</u>      |
| 3. B(CO <sub>3</sub> ) <sub>3</sub>                | <u>boron carbonate</u>     |
| 4. Mg(OH) <sub>2</sub>                             | <u>magnesium hydroxide</u> |
| 5. MgF <sub>2</sub>                                | <u>magnesium fluoride</u>  |
| 6. (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> | <u>ammonium sulfate</u>    |
| 7. K <sub>3</sub> N                                | <u>potassium nitride</u>   |
| 8. LiNO <sub>3</sub>                               | <u>lithium nitrate</u>     |
| 9. BaSO <sub>4</sub>                               | <u>barium sulfate</u>      |
| 10. Mg(NO <sub>3</sub> ) <sub>2</sub>              | <u>magnesium nitrate</u>   |
| 11. AgCl   | <u>silver chloride</u>     |
| 12. Al(OH) <sub>3</sub>                            | <u>aluminum hydroxide</u>  |
| 13. CaSO <sub>4</sub>                              | <u>calcium sulfate</u>     |
| 14. Ca <sub>3</sub> P <sub>2</sub>                 | <u>calcium phosphide</u>   |
| 15. SrS  | <u>strontium sulfide</u>   |
| 16. BeCl <sub>2</sub>                              | <u>beryllium chloride</u>  |
| 17. NaI  | <u>sodium iodide</u>       |
| 18. AlCl <sub>3</sub>                              | <u>aluminum chloride</u>   |
| 19. MgCO <sub>3</sub>                              | <u>magnesium carbonate</u> |

# Energy Section 3 Reinforcement Answers

**Trevor Letcher**



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