## Part A: Periodic Table Webquest

#### Click on the first website listed http://www.chem4kids.com/files/elem\_pertable.html

	All and the company of the company o		
	A row is also called a		
3.	All of the elements found within		
4.			
	Columns are also called		
6.		ve the same number of	which are invol-
	in		
	Each element within group 1 ha		
8.			in its outer shell but it is st
	called a	because its outer orbit is	
On the	e left hand side under Element Gr	oups click Alkali Metals.	
	on the second website - http://v		
9.	They are found in group		
9.	They are found in group		Why do y
9.	They are found in group	and do not occur	Why do y
9. 10	They are found in group	and do not occur	
9. 10	They are found in group  They are very think this is? How many electrons do they los	and do not occur se when bonding and what kind	
9. 10 11	They are found in group  They are very think this is?  How many electrons do they lose	and do not occur se when bonding and what kind	
9. 10 11	They are found in group  They are very think this is?  How many electrons do they lose	and do not occur se when bonding and what kinds	s of bonds do they form?
9. 10 11 12 13	They are found in group They are very think this is? How many electrons do they local the control of the	and do not occurse when bonding and what kinds	s of bonds do they form? Why do they explode? You wi
9. 10 11 12 13	They are found in group  They are very think this is?  How many electrons do they located to the state of the state	and do not occur se when bonding and what kinds ls will ment Groups click Alkaline Eart	s of bonds do they form? Why do they explode? You with Metals.
9. 10 11 12 13	They are found in group think this is? How many electrons do they lose. Like most metals they are If exposed to water, alkali metal need to do additional research On the left hand side under Electrons are These metals are	and do not occur se when bonding and what kinds ls will ment Groups click Alkaline Eart They are also	s of bonds do they form? Why do they explode? You with Metals.
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**Mark Steyvers** 



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