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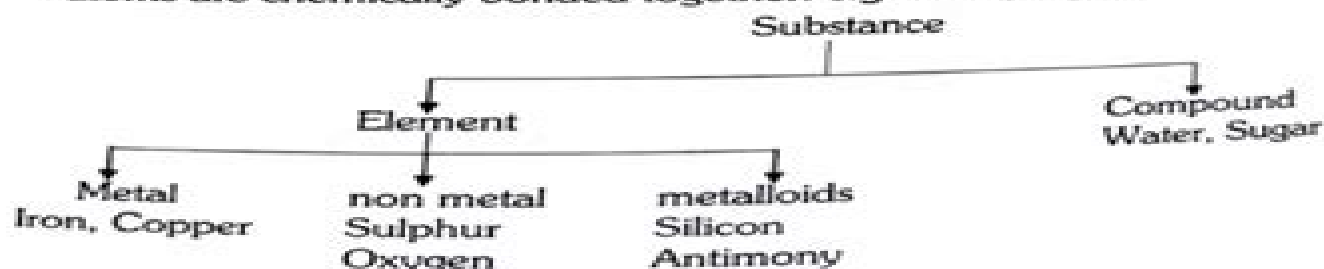
Chapter- Metals and non Metals,

Some Basic Concepts

1. Pure substance- Pure substance can either be made of same type of atoms or they may be made up of two or more types of atoms.

Pure substance are of two types

- (a) Elements- A substance that is made up of only one kind of atoms and cannot be broken further into smaller components is called an element.
Element is further classified as metal, non metals and metalloids. e.g Iron, Copper.
- (b) Compounds- A compound is a substance formed when 2 or more different atoms are chemically bonded together. e.g Water, sugar.



(Class - VIII : SCIENCE)

Common elements and their symbols

English name of the element		Symbol	English name of the element		Symbol
1.	Argon	Ar	15.	Iodine	I
2.	Arsenic	As	16.	Lithium	Li
3.	Aluminium	Al	17.	Magnesium	Mg
4.	Boron	B	18.	Manganese	Mn
5.	Barium	Ba	19.	Nitrogen	N
6.	Bromine	Br	20.	Neon	Ne
7.	Carbon	C	21.	Nickel	Ni
8.	Calcium	Ca	22.	Oxygen	O
9.	Chlorine	Cl	23.	Phosphorus	P
10.	Chromium	Cr	24.	Platinum	Pt
11.	Cobalt	Co	25.	Sulphur	S
12.	Fluorine	F	26.	Uranium	U
13.	Hydrogen	H	27.	Zinc	Zn
14.	Helium	He			

TABLE 2. Symbols derived from Latin names

Metals Section Ch 19 Answers

A Loxley



Metals Section Ch 19 Answers:

Reviewing **Metals Section Ch 19 Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Metals Section Ch 19 Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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