

Lewis dot symbols of the atoms of elements for the main groups in the periodic table

Group No

Period No

s-Block

d-Block

p-Block

1	2						
Alkali metals	Alkaline earth metals						
Valence electrons 1	Valence electrons 1						
Hydrogen 1 • H 1s ¹							
Lithium 3 • Li [He] 2s ¹	Beryllium 4 • Be [He] 2s ²						
Sodium 11 • Na [Ne] 3s ¹	Magnesium 12 • Mg [Ne] 3s ²						
Potassium 19 • K [Ar] 4s ¹	Calcium 20 • Ca [Ar] 4s ²						
Rubidium 37 • Rb [Kr] 5s ¹	Strontium 38 • Sr [Kr] 5s ²						
Cesium 55 • Cs [Xe] 6s ¹	Barium 56 • Ba [Xe] 6s ²						
Francium 87 • Fr [Ra] 7s ¹	Radium 88 • Ra [Ra] 7s ²						
		3	4	5	6	7	8
		Boron family	Carbon family	Nitrogen family	Oxygen family	Halogens	Noble gases
		Valence electrons 3	Valence electrons 4	Valence electrons 5	Valence electrons 6	Valence electrons 7	Valence electrons 8
							Helium 2 • He 1s ²
		Boron 5 • B [He] 2s ² 2p ¹	Carbon 6 • C [He] 2s ² 2p ²	Nitrogen 7 • N [He] 2s ² 2p ³	Oxygen 8 • O [He] 2s ² 2p ⁴	Fluorine 9 • F [He] 2s ² 2p ⁵	Neon 10 • Ne [He] 2s ² 2p ⁶
		Aluminium 13 • Al [Ne] 3s ² 3p ¹	Silicon 14 • Si [Ne] 3s ² 3p ²	Phosphorus 15 • P [Ne] 3s ² 3p ³	Sulphur 16 • S [Ne] 3s ² 3p ⁴	Chlorine 17 • Cl [Ne] 3s ² 3p ⁵	Argon 18 • Ar [Ne] 3s ² 3p ⁶
		Gallium 31 • Ga [Ar] 3d ¹⁰ 4s ¹ 4p ¹	Germanium 32 • Ge [Ar] 3d ¹⁰ 4s ² 4p ²	Arsenic 33 • As [Ar] 3d ¹⁰ 4s ² 4p ³	Selenium 34 • Se [Ar] 3d ¹⁰ 4s ² 4p ⁴	Bromine 35 • Br [Ar] 3d ¹⁰ 4s ² 4p ⁵	Krypton 36 • Kr [Ar] 3d ¹⁰ 4s ² 4p ⁶
		Indium 49 • In [Kr] 4d ¹⁰ 5s ¹ 5p ¹	Tin 50 • Sn [Kr] 4d ¹⁰ 5s ² 5p ²	Antimony 51 • Sb [Kr] 4d ¹⁰ 5s ² 5p ³	Tellurium 52 • Te [Kr] 4d ¹⁰ 5s ² 5p ⁴	Iodine 53 • I [Kr] 4d ¹⁰ 5s ² 5p ⁵	Xenon 54 • Xe [Kr] 4d ¹⁰ 5s ² 5p ⁶
		Thallium 81 • Tl [Xe] 4f ¹⁴ 5d ¹⁰ 6s ¹ 6p ¹	Lead 82 • Pb [Xe] 4f ¹⁴ 5d ¹⁰ 6s ² 6p ²	Bismuth 83 • Bi [Xe] 4f ¹⁴ 5d ¹⁰ 6s ² 6p ³	Polonium 84 • Po [Xe] 4f ¹⁴ 5d ¹⁰ 6s ² 6p ⁴	Astatine 85 • At [Xe] 4f ¹⁴ 5d ¹⁰ 6s ² 6p ⁵	Radon 86 • Rn [Xe] 4f ¹⁴ 5d ¹⁰ 6s ² 6p ⁶
		Uubium 112 • Uu [Xe] 5f ¹⁴ 6d ¹⁰ 7s ² 7p ²	Flerovium 114 • Fl [Xe] 5f ¹⁴ 6d ¹⁰ 7s ² 7p ²	Uupium 115 • Uu [Xe] 5f ¹⁴ 6d ¹⁰ 7s ² 7p ³	Livermorium 116 • Lv [Xe] 5f ¹⁴ 6d ¹⁰ 7s ² 7p ⁴	Oganeson 117 • Og [Xe] 5f ¹⁴ 6d ¹⁰ 7s ² 7p ⁵	Uuoium 118 • Uu [Xe] 5f ¹⁴ 6d ¹⁰ 7s ² 7p ⁶

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Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance *BIOS Instant Notes in Chemistry for Biologists* Julie

Fisher, John Arnold, 2003-09-25 Instant Notes in Chemistry for Biologists is a concise book for undergraduates who have a limited background in chemistry This book covers the main concepts in chemistry provides simple explanations of chemical terminology and illustrates underlying principles and phenomena in the life sciences with clear biological examples Building on the success of the first edition the second edition has been fully revised and updated and comprises new sections on water as a biological solvent inorganic molecules and biological macromolecules *BIOS Instant Notes in Inorganic Chemistry*

Tony Cox, 2004-03-01 Instant Notes in Inorganic Chemistry second edition has been fully updated and new material added on

developments in noble gas chemistry and the synthesis reactions and characterization of inorganic compounds New chapters cover the classification of inorganic reaction types concentrating on those useful in synthesis techniques used in characterizing compounds including elemental analysis spectroscopic methods IR NMR and structure determination by X ray crystallography and the factors involved in choosing appropriate solvents for synthetic reactions The new edition continues to provide concise coverage of inorganic chemistry at an undergraduate level offering easy access to all important areas of inorganic chemistry in a format which is ideal for learning and rapid revision

Understanding General Chemistry Atef Korchef, 2022-03-07 Understanding General Chemistry details the fundamentals of general chemistry through a wide range of topics relating the structure of atoms and molecules to the properties of matter Written in an easy to understand format with helpful pedagogy to fuel learning the book features main objectives at the beginning of each chapter get smart sections and check your reading section at the end of each chapter The text is filled with examples and practices that illustrate the concepts at hand In addition a summary and extensive MCQs exercises and problems with the corresponding answers and explanations are readily available Additional features include Alerts students to common mistakes and explains in simple ways and clear applications how to avoid these mistakes Offers answers and comments alongside sample problems enabling students to self evaluate their skill level Includes powerful methods easy steps simple and accurate interpretations and engaging applications to help students understand complex principles Provides a bridge to more complex topics such as solid state chemistry organometallic chemistry chemistry of main group elements inorganic chemistry and physical chemistry This introductory textbook is ideal for chemistry courses for non science majors as well as health sciences and preparatory engineering students

Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice) Heather Hattori, Richard H. Langley, 2014-04-14 Practice makes perfect and helps deepen your understanding of chemistry Every high school requires a course in chemistry and many universities require the course for majors in medicine engineering biology and various other sciences 1001 Chemistry Practice Problems For Dummies provides students of this popular course the chance to practice what they learn in class deepening their understanding of the material and allowing for supplemental explanation of difficult topics 1001 Chemistry Practice Problems For Dummies takes you beyond the instruction and guidance offered in Chemistry For Dummies giving you 1 001 opportunities to practice solving problems from the major topics in chemistry Plus an online component provides you with a collection of chemistry problems presented in multiple choice format to further help you test your skills as you go Gives you a chance to practice and reinforce the skills you learn in chemistry class Helps you refine your understanding of chemistry Practice problems with answer explanations that detail every step of every problem Whether you re studying chemistry at the high school college or graduate level the practice problems in 1001 Chemistry Practice Problems For Dummies range in areas of difficulty and style providing you with the practice help you need to score high at exam time

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Fisher, J.R.P. Arnold, Julie Fisher, John Arnold, 2020-09-15 BIOS Instant Notes Chemistry for Biologists Third Edition is the perfect text for undergraduates looking for a concise introduction to the subject or a study guide to use before examinations. Each topic begins with a summary of essential facts, an ideal revision checklist followed by a description of the subject that focuses on core information with clear simple diagrams that are easy for students to understand and recall in essays and exams. BIOS Instant Notes Chemistry for Biologists Third Edition is fully up to date and covers: The elements, Chemical bonds and molecular shape, Water the biological solvent, Carbon the basis for life on Earth, 3D molecular structure of organic compounds, Small inorganic molecules of biological importance, Some metals in biology, Molecular interactions, Common reaction types of carbon based compounds, Organic compounds by chemical class, Aromatic compounds, Chemical synthesis of biological molecules, Important biological macromolecules by class, Aqueous behaviour, Elementary thermodynamics, Kinetics, Spectroscopy, Units and calculations.

Chemistry for the IB Diploma Workbook with CD-ROM Jacqueline Paris, 2017-04-06 Chemistry for the IB Diploma Second edition covers in full the requirements of the IB syllabus for Chemistry for first examination in 2016. This workbook is specifically for the IB Chemistry syllabus for examination from 2016. The Chemistry for the IB Diploma Workbook contains straightforward chapters that build learning in a gradual way, first outlining key terms and then providing students with plenty of practice questions to apply their knowledge. Each chapter concludes with exam style questions. This structured approach reinforces learning and actively builds students confidence using key scientific skills: handling data, evaluating information and problem solving. This helps empower students to become confident and independent learners. Answers to all of the questions are on the CD ROM.

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