

(vi) What is the IUPAC name of



?

(a) α - Amino butanoic acid

(b) 2 - Amino butyric acid

(c) α - Amino butyric acid

(d) 2 - Amino butanoic acid

(vii) Which among the following molecular formulae represents urotropine?

(a) $C_6H_{12}N_4$

(b) $C_6H_{24}H_4$

(c) $C_6H_{12}N_4O_2$

(d) $C_6H_{24}N_4O_2$

Q. 6. Answer any THREE of the following:

[12]

(i) Write the structures of –

(a) 3-chloro-3-ethylhex-1-ene

(b) 1-Iodo-2, 3-dimethylbutane

(c) 1, 3, 5 – tribromobenzene

(ii) What is the action of acidified potassium dichromate on – (a) SO_2 (b) KI

Draw structure of dichromate ion.

(iii) Describe laboratory method for preparation of glucose.

Write the reaction that indicates the presence of $-CHO$ group in glucose.

(iv) What will be the action of the mixture of sodium nitrite and dilute hydrochloric acid on – (a) ethyl amine (b) aniline (c) triethyl amine

Q. 7. Answer any SIX of the following :

[9]

(i) What are chemical twins? Write 'two' examples.

(ii) Explain the terms : (a) Antiseptics (b) Analgesics

(iii) Draw the simple Fisher projection formulae of D - (+) - glucose and D - (-) - fructose.

(iv) Classify the following ligands into monodentate and polydentate –

(a) Ammonia

(b) Carbon monoxide

(c) Ethylene diamine

(d) Ethylene diamine tetra acetate ion

(v) State and explain Markownikoff's rule with suitable example.

(vi) How are propan-1-amine and propan-2-amine prepared from oxime?

(vii) Identify 'A' and 'B' in the following reaction :



(viii) What is the action of the following reagents on phenol –

(a) Bromine in CS_2 at low temperature.

(b) Conc. H_2SO_4 at room temperature.

Q. 8. Answer any ONE of the following:

[7]

(i) Write the structure and IUPAC names of all the metamers represented by formula $C_4H_{10}O$.

Write balanced chemical equations for action of ammonia on –

(a) formaldehyde

(b) acetaldehyde

(c) acetone

(ii) Write 'four' characteristics of co-ordinate complex ion.

How is Nylon 6, 6 prepared?

Write any 'two' uses of terylene.

Write 'four' physical methods of preserving food materials.

Hsc Chemistry Board Paper October 2014

H Kauffman



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