

ICSE Class 10 Chemistry Question Paper Solution 2017

CHEMISTRY (SCIENCE PAPER-2)

SECTION I (40 Marks)

Attempt all questions from this Section

Question 1

- (a) Fill in the blanks from the choices given in brackets: [5]
- (i) The energy required to remove an electron from a neutral isolated gaseous atom and convert it into a positively charged gaseous ion is called _____.
(*electron affinity, ionisation potential, electronegativity*)
 - (ii) The compound that does not have a lone pair of electrons is _____.
(*water, ammonia, carbon tetra chloride*)
 - (iii) When a metallic oxide is dissolved in water, the solution formed has a high concentration of _____ ions. (*H⁺, H₃O⁺, OH⁻*)
 - (iv) Potassium sulphite on reacting with hydrochloric acid releases _____.
gas. (*Cl₂, SO₂, H₂S*)
 - (v) The compound formed when ethene reacts with Hydrogen is _____.
(*CH₄, C₂H₆, C₃H₈*)
- (b) Choose the **correct answer** from the options given below: [5]
- (i) A **chloride** which forms a precipitate that is soluble in excess of ammonium hydroxide, is:
 - 1. Calcium chloride
 - 2. Ferrous chloride
 - 3. Ferric chloride
 - 4. Copper chloride
 - (ii) If the molecular formula of an organic compound is **C₁₆H₁₈** it is:
 - 1. alkene

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Table of Contents Icse Chemistry Question Paper

1. Understanding the eBook Icse Chemistry Question Paper
 - The Rise of Digital Reading Icse Chemistry Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Icse Chemistry Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Icse Chemistry Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Icse Chemistry Question Paper
 - Personalized Recommendations
 - Icse Chemistry Question Paper User Reviews and Ratings
 - Icse Chemistry Question Paper and Bestseller Lists
5. Accessing Icse Chemistry Question Paper Free and Paid eBooks

- Icse Chemistry Question Paper Public Domain eBooks
- Icse Chemistry Question Paper eBook Subscription Services
- Icse Chemistry Question Paper Budget-Friendly Options
- 6. Navigating Icse Chemistry Question Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Icse Chemistry Question Paper Compatibility with Devices
 - Icse Chemistry Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Icse Chemistry Question Paper
 - Highlighting and Note-Taking Icse Chemistry Question Paper
 - Interactive Elements Icse Chemistry Question Paper
- 8. Staying Engaged with Icse Chemistry Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Icse Chemistry Question Paper
- 9. Balancing eBooks and Physical Books Icse Chemistry Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Icse Chemistry Question Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Icse Chemistry Question Paper
 - Setting Reading Goals Icse Chemistry Question Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Icse Chemistry Question Paper
 - Fact-Checking eBook Content of Icse Chemistry Question Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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