

D. Presence of a Catalyst

- What effect does the MnO_2 catalyst have on the rate of evolution of O_2 gas?
- Write a balanced equation for the decomposition of H_2O_2 .

E. Concentration of Reactants: Magnesium-Hydrochloric Acid System

Concentration of HCl	Volume of HCl	mol HCl	mass of Mg	mol Mg	mol HCl / mol Mg	Time (sec)
6 M	5 mL		0.0301			
4 M	5 mL		0.0188			
3 M	5 mL		0.0161			
1 M	5 mL		0.0124			

- Plot $\frac{\text{mol HCl}}{\text{mol Mg}}$ (y-axis) versus time (x-axis). Instructor's approval of graph: _____
- From your graph, predict the time, in seconds, for 5 mg of Mg to react in 5 mL of 2.0 M HCl.

F. Concentration of Reactants: Iodic Acid-Sulfurous Acid System

Molar concentration of HIO_3 _____ Molar concentration of H_2SO_3 _____

- What is the volume (mL) per drop of the HIO_3 solution? _____ What is the volume (mL) per drop of the water? _____

Test Tube	Drops of HIO_3	mL HIO_3	Drops of H_2O	mL H_2O	mL H_2SO_3	$[\text{HIO}_3]_0$ (diluted)	Time (sec)
1					1.0		
2					1.0		
3					1.0		
4					1.0		
5					1.0		

- Plot $[\text{HIO}_3]_0$ (y-axis) versus time (x-axis). Instructor's approval of graph: _____
- How does a change in the molar concentration of HIO_3 affect the time required for the appearance of the deep-blue I_2 -starch complex?
- Estimate the time, in seconds, for the deep-blue I_2 -starch complex to form when 10 drops of 0.01 M HIO_3 are used for the reaction. Assume all other conditions remain constant.

^aSee footnote 5.

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Albert A Gayle



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