

Thermodynamics

Section Quiz: The First Law of Thermodynamics

Write the letter of the correct answer in the space provided.

- _____ 1. Which concept does the first law of thermodynamics describe?
 - a. conservation of mass
 - b. conservation of energy
 - c. energy being transformed
 - d. conservation of momentum
- _____ 2. Which concept allows $Q = 0$ in that the first law of thermodynamics takes the form of $\Delta U = -W$?
 - a. an adiabatic process
 - b. an isothermal process
 - c. an isobaric process
 - d. an isochoric process
- _____ 3. Which concept allows $W = 0$ in that the first law of thermodynamics takes the form of $\Delta U = Q$?
 - a. an adiabatic process
 - b. an isothermal process
 - c. an isobaric process
 - d. an isochoric process
- _____ 4. Which process is isothermal, $Q = W = 0$, in that the first law of thermodynamics takes the form of $\Delta U = 0$?
 - a. an adiabatic process
 - b. an isothermal process
 - c. an isobaric process
 - d. an isochoric process
- _____ 5. Which concept allows $\Delta U = 0$ in that the first law of thermodynamics takes the form of $Q = W$?
 - a. an adiabatic process
 - b. an isothermal process
 - c. an isobaric process
 - d. an isochoric process
- _____ 6. A process requires 2000 J of energy to heat a body, giving up 3000 J of energy to the body's surroundings, and loses 1000 J of energy. What is the change in the internal energy of the body?
 - a. 1000 J
 - b. 2000 J
 - c. 3000 J
 - d. 4000 J

Holt Physical Science Quiz Section Temperature Answers

Annelies Wilder-Smith



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