

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

Holt Modern Chemistry Homework Answers Reviewing Concepts

Ralph Thomas Myers



Holt Modern Chemistry Homework Answers Reviewing Concepts:

Unveiling the Energy of Verbal Beauty: An Mental Sojourn through **Holt Modern Chemistry Homework Answers Reviewing Concepts**

In some sort of inundated with monitors and the cacophony of immediate conversation, the profound energy and emotional resonance of verbal artistry often disappear into obscurity, eclipsed by the constant assault of sound and distractions. Yet, situated within the musical pages of **Holt Modern Chemistry Homework Answers Reviewing Concepts**, a interesting perform of fictional elegance that impulses with fresh thoughts, lies an wonderful trip waiting to be embarked upon. Penned by a virtuoso wordsmith, that interesting opus books viewers on an emotional odyssey, lightly exposing the latent possible and profound influence stuck within the intricate web of language. Within the heart-wrenching expanse of this evocative examination, we shall embark upon an introspective exploration of the book is main subjects, dissect their captivating publishing style, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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