#### Name the following acids

- 1. HCL
- 2. HClO<sub>4</sub>
- HIO<sub>3</sub>
- 4. HII
- 5. H<sub>2</sub>SO<sub>4</sub>
- 6. H<sub>2</sub>S
- 7. HCN
- 8. H<sub>2</sub>CO<sub>3</sub>
- 9. HBrO4
- 10.HBrOs
- 11.HC2H3O2
- 12.H<sub>3</sub>PO<sub>3</sub>
- 13.H<sub>3</sub>P
- 14.H<sub>2</sub>CrO<sub>4</sub>
- 15.H<sub>2</sub>CrO<sub>2</sub>
- 16.H2Cr

#### Give formulas for the following acids

- 17. hydrofluoric acid
- 18 nitric acid
- 19.nitrous acid
- 20.hydroxic acid
- 21 periodic acid
- 22.percarbonic acid
- 23.hypobromous acid
- 24. bromous acid
- 25.permanganic acid
- 26.manganic acid
- 27. iodous acid
- 28. sulfurous acid
- 29. sulfuric acid
- 30.perchloric acid
- 31.chlorous acid
- 32.acetic acid
- 33.phosphoric acid
- 34.phosphorous acid

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Yan Bai

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