

Griddable Science Items

Topography map

Density

Atoms

$$F = ma$$

			6	.	2	5
0	0	0	0		0	0
1	1	1	1		1	1
2	2	2	2		2	2
3	3	3	3		3	3
4	4	4	4		4	4
5	5	5	5		5	5
6	6	6	6		6	6
7	7	7	7		7	7
8	8	8	8		8	8
9	9	9	9		9	9

Designed for

8th grade STAAR Test



but would work for other standardized tests


Cricket Science

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1. A balance and a graduated cylinder are used to determine the density of a mineral sample. The sample has a mass of 16.7 g and a volume of 14.2 cm³. What is the density of the mineral sample in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{16.7}{14.2} = 1.18 \text{ g/cm}^3$$

2. 4 mL is 4 cm³.
What is the density of the rock in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{18.00}{4} = 4.50 \text{ g/cm}^3$$

3. Measure your classmates' hand for the problems. Use your hand to use the


1. 10 cm 20 cm 30 cm 40 cm 50 cm 60 cm 70 cm 80 cm 90 cm 100 cm

2. 10 cm 20 cm 30 cm 40 cm 50 cm 60 cm 70 cm 80 cm 90 cm 100 cm

3. 10 cm 20 cm 30 cm 40 cm 50 cm 60 cm 70 cm 80 cm 90 cm 100 cm


Name: _____ ANSWERS

1. What is the density of the rock in the graduated cylinder?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{4.00}{2} = 2.00 \text{ g/cm}^3$$

4. The sample of metal has a mass of 35 g. What is the density in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{35}{10} = 3.5 \text{ g/cm}^3$$

Design

81

ST

Gridable Science 8th Grade Staar Practice

Ensheng Dong



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