Metals versus Non-Metals

Metals are on the left side. Non-metals on the right. Metals tend to lose electrons. Non-metals gain them tight.

The dividing line is the red or bold black line from between Boron and Alaminum down and to the right. Everything to the left is a metal: to the right, mon-metal. One exception (don't ya just bate that?!) is hydrogen — a non-metal.

Sodium (Nn) is a metal. Oxygen is a non-metal.

Metal or Non-metal?				
Potassium:	Fluorine:			
Brownine:	Hydrogen			
Beryllium:	Silver			
Heliam:	Nitrogen			

1		Dividing line Non-Metals						
35 t. in	4 Be		5 15	6 C	7 N	8 O	9 F	100 300g
Nin	1.2		13	1.4	15	8-6	17	1.8
Nin	Mg		Al	Si	P	S	CI	A.r
191	20		31	32	33	34	36	3.6
K	Ca		Ga	Ge	As	Se	Be	Ka

Metals

Easy to remember by which side iron (Fe) is on—the left side! On either side of the divide are the metalloids or semi-metals they have characteristics of both B. St. Ge, As.

Dot Diagrams

Dot Diagrams (sometimes known as Lewis dot diagrams) are a depiction of an atom's valence electrons. They are a powerful tool in helping you understand, see, and even predict molecular bonding.

Put the correct number of <u>valence</u> electrons around the chemical symbol in this order in pairs:



Neon has 8 valence electrons (in the right hand column). Neon has no unoccupied spaces. It is full. Notice that the electrons are in pairs not a circle.

Why would the electrons spread out around the atom instead of bunching up? Li

Lithiam, with I valence electron (first column).

٠Ċ٠

Carbon has 4 volence electrons



Oxygen has 6 valence electrons. How many more before it's full?

Draw the Dot Diagrams for the following elements:

Beryllium

Sodium

Helican

Carbon-

Borom

<u>Metals Vs Nonmetals Dot Diagrams Ions Worksheet</u> <u>Answer Key</u>

Robert R. Redfield

Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key:

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Table of Contents Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key

- 1. Understanding the eBook Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - The Rise of Digital Reading Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Personalized Recommendations
 - Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key User Reviews and Ratings

- Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key and Bestseller Lists
- 5. Accessing Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key Free and Paid eBooks
 - Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key Public Domain eBooks
 - Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key eBook Subscription Services
 - Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key Budget-Friendly Options
- 6. Navigating Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key Compatibility with Devices
 - Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Highlighting and Note-Taking Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Interactive Elements Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
- 8. Staying Engaged with Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
- 9. Balancing eBooks and Physical Books Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Setting Reading Goals Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Fact-Checking eBook Content of Metals Vs Nonmetals Dot Diagrams Ions Worksheet Answer Key
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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