



IONIC AND METALLIC BONDING

Practice Problems

Do your best to solve these problems on your own.

SECTION 7.1 IONS

1. How many electrons does each atom have? How many protons does each atom have? How many electrons does each ion have? How many protons does each ion have?

a. Na^+ b. Cl^- c. Ca^{2+}

2. How many electrons does each atom have? How many protons does each atom have?

a. K^+ b. Br^- c. Al^{3+}

3. How many electrons does each atom have? How many protons does each atom have?

a. Fe^{2+} b. S^{2-} c. Mg^{2+} d. Zn^{2+} e. P^{3-}

4. How many electrons does each atom have? How many protons does each atom have? How many electrons does each ion have? How many protons does each ion have?

5. How many electrons does each atom have? How many protons does each atom have? How many electrons does each ion have? How many protons does each ion have?

a. Na^+ b. Cl^- c. Ca^{2+} d. K^+ e. Br^- f. Al^{3+}

6. How many electrons does each atom have? How many protons does each atom have?

a. Na^+ b. Cl^- c. Ca^{2+} d. K^+

7. How many electrons does each atom have? How many protons does each atom have? How many electrons does each ion have? How many protons does each ion have?

a. Fe^{2+} b. S^{2-} c. Mg^{2+} d. Zn^{2+}

8. How many electrons does each atom have? How many protons does each atom have?

a. Fe^{2+} b. S^{2-} c. Mg^{2+} d. Zn^{2+} e. P^{3-} f. N^{3-}

Ionic And Metallic Bonding Answer Key Chapter Test A

Michal Rosen-Zvi



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