Name:	KEY	Block:

Assignment #2 - Compound Names and Formulas Multi-valent ions only

A. Write the correct formula for the	e following compounds, all of which have been named using the
modern Roman Numeral Method.	The ionic charge is given after the first element (metallic) in Roman
Numerals.	
C C	

1. copper (II) oxide	11. manganese (III) oxide MY 203		
2. mercury (I) oxide	12. vanadium (II) bromide VB(z		
3. gold (III) chloride AuCl3	13. niobium(V) oxide Nb205		
4. thallium (III) bromide TIBY 3	14. titanium (III) oxide		
5. bismuth (V) oxide <u>B1205</u>	15. titanium (III) nitride TIN		
6. terbium (III) oxideT\oz0_3	16. iron (II) oxide FCO		
7. uranium (VI) oxide \\(\lambda \)	17. cobalt (II) phosphide C03P2		
8. protactinium (IV) bromide PhBY4	18. tín (II) oxide		
9. cerium (III) oxide	19. thulium (II) bromide TWB(2		
10. arsenic (V) sulphide A52SS	20. copper (II) bromide CUBY2		
B. Name the compound using the Roman Numeral Method and balance the ionic charges.			
1. Snck _ Tin (IV) Chloride	9. Pdf. Ralladium (IV) Fluoride		
2. BIBro Bismuth (V) Bromide	10. Os2O3 OSMIUM (III) Oxide		
3. POO2 POLONIUM (IV) OXICE	11. MOBEZ MOLY/Odenium (II) Bramide		
4. Pbl2 100d(II) Iodide	12. VCI6 Vanadium (V) Chloride		
5. HgO Mercury (II) Oxide	13. Mn2O3 Managamese (III) Oxide		
6. HgCI _ MEYCUYY (I) Chroride_	14. coo _ COLODIA (II) Oxide		
7. Augos Gold (III) Oxide.	15. NP2O3 NEDTWOMM (III) OXIC		
8. FeCla Iron (II) Chicyide	16. V2Os Vanadium (V) Oxide		

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