Draw the Lewis structure for water, H₂O.

- a) How many "groups" (atoms and lone pairs) surround the central oxygen?
- b) What is the geometry of this molecule (look at atoms and lone pairs)? Draw this VSEPR structure next to the Lewis structure.
- c) What is the shape of this molecule (look only at the atoms)?
- d) What is the H-O-H bond angle?
- e) Place the partial positive and negative charges on the H and O atoms, based on their relative electronegativities. Is water a polar compound?

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contain two short artifact reports The papers focus on aspects of programming language research such as programming paradigns and styles methods and tools to specify and reason about programs and languages programming language foundations methods and tools for implementation concurrency and districution and applications and emerging topics

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