

## Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	very	very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces $H_2$ gas	—
Reaction with Carbonate	produces $CO_2$ gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
$HNO_3$	A	nitric acid
$Mg(OH)_2$	B	magnesium hydroxide
$H_3PO_4$	A	phosphoric acid
$H_2CO_3$	A	carbonic acid
HF	A	hydrofluoric acid
$Fe(OH)_3$	B	iron (iii) hydroxide

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u><math>Al(OH)_3</math></u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u><math>Ca(OH)_2</math></u>
sulfuric acid	<u><math>H_2SO_4</math></u>	sulfurous acid	<u><math>H_2SO_3</math></u>

# Intro To Acids And Bases Worksheet Answers

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address students common misconceptions Detailed descriptions of how these instructional approaches can be incorporated into teaching and learning science are also included The science education literature extensively documents the findings of studies about students misconceptions or alternative conceptions about various science concepts Furthermore some of the studies involve systematic approaches to not only creating but also implementing instructional programs to reduce the incidence of these misconceptions among high school science students These studies however are largely unavailable to classroom practitioners partly because they are usually found in various science education journals that teachers have no time to refer to or are not readily available to them In response this book offers an essential and easily accessible guide

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