Name: KEY _____ Solubility WS

Use your solubility curve chart to answer the following questions.

Part 1: Graph interpretation: (no work needed)

What are the customary units of solubility on solubility curves? Grams of Solute per 100 g of water

- According to the graph, the solubility of any substance changes as temperature changes.
- Most substances on this graph show increased solubility as temperature increases. What are the exceptions? Ce₂(SO₄)₃ and NH₃
- The solubilities of substances whose curves show greater (steeper) slopes are
- (more/less) affected by temperature changes than those that have more gradual slopes.
- Which salt has solubility values that are least affected by changes in temperature? NaCI
- What is the solubility of ammonium chloride at 60°C? 50 g/100 g H₂O
- At what temperature do potassium chlorate and potassium chloride have the same solubility in water? 95 °C
- Which compound is least soluble in water at 12 °C? KClO₃
- A saturated solution of which compound contains 130 grams of solute per 100 grams of water at 70°C? KNO₃
- Are the following solutions unsaturated, saturated, or supersaturated?
 80 g of sodium nitrate in 100 g of water at 30℃. unsaturated
 80 g of potassium chlorate in 100 g of water at 50℃. supersaturated
- 10. Which saturated solution of a chloride has the greatest percentage by mass of solute at 60°C? don't worry about this question for the test

Problems: Show all work on a separate sheet.

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