

# 9 • Bonding & Molecular Structure

## PRACTICE TEST

1. The correct Lewis symbol for ground state carbon



2. The correct Lewis symbol for ground state aluminum is



3. Using the picture below, what process corresponds to the lattice energy?



4. Which of the following favors formation of an ionic compound?

- (a) low ionization energy for metal  
(b) high electron affinity for non metal  
(c) high lattice energy  
(d) all of a-c above  
e) none of a-c above

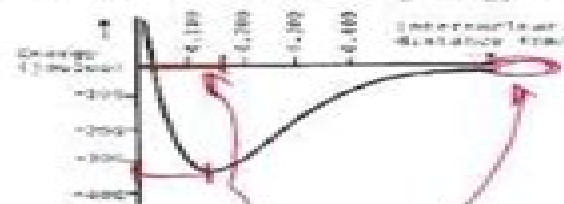
5. Which of the atoms below is least likely to violate the octet rule?



6. How many electrons are shown in the Lewis structure of perchlorate ion,  $\text{ClO}_4^-$ ?



- Questions 7 - 9 refer to the following energy diagram:



7. What is the energy of the two isolated atoms?

- a) -400 J d) 0.100 nm  
b) -335 J e) -155 J  
(c) 0 J

8. What is the bond length of the bond between the two atoms?

- a) 0.020 nm d) 0.330 nm  
(b) 0.140 nm e) 2.00 nm  
c) 0.400 nm

9. What is the bond energy (bond strength) of the bond?

- a) 0 J (d) -330 J Should this be + value? Unknown  
b) -110 J e) -422 J  
c) -10 J

10. What is the bond order of the C-O bond in acetone?



- a) 4 b) 1.5 c) 0.5 d) 1 (e) 2

# Larson Geometry Chapter Two Test Answers

**Joacim Rocklöv**



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In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Larson Geometry Chapter Two Test Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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