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2021 MSCE RESULTS

NO	SUR NAME	FIRST NAME	AGRIC.	BIO	BUS	CHEM	CHIC	COMPU.	ENGLISH	FRENCH	GEOGRA	HISTORY	MATHS	ADD. MA	PHYSICS	AGGR.
1	GOMBACHIKA	SIBUSISO MARK	2	2		2	7	2	3		2		1	1	2	10
2	VINCENT	VINTHENG	3	3	3	2	6		3		1		1	1	2	10
3	CHAYEKHA	MURHENDERE MIGUEL	2	2	2	2	5		1		2	2	2		3	11
4	MAPANJE	NOEL	2	3		2	4	1	2		2		2	2	4	11
5	MWALILINO	PROMISE	2	2		3	5	2	5		2		2	1	3	11
6	MWENEBANDA	BENTRY PETER	2	3	2	2	5		3		2		2	1	3	11
7	MZIKAMANDA	CHIMWEMWE	2	3	2	2	5		2		1	2	5		4	11
8	CHALE	MICHAEL DAVID	2	3	2	2	4		2		2		2	2	3	12
9	CHIKOPA	INNOCENT BRAVE JN	2	3	2	2	4		2		2	2	3		4	12
10	CHINYAMU	WISDOM	3	2		2	6	2	3		2		2		2	12
11	CHITEKWE	TAKONDWA JONATH	2	3		2	3	2	2	4	2		3	2	3	12
12	KAKOTA	MADALITSO MARTIN	2	2		2	4	2	3		2	2	4		3	12
13	KALONGA	AKUZIKE	2	2	3	2	7		2		2	2	2		2	12
14	KUYOKWA	MTENDE JOSEPH	2	3	2	2	5		4	6	2		2		2	12
15	MASOKA	HAROLD	2	2		2	6	3	4		2		2	2	2	12
16	MLENDU	MARVEL	2	3	2	2	4		3		2	2	2		3	12
17	MTONGA	VITUMBIKO	2	2		2	4	2	2		2		2	2	2	12
18	MZUMARA	ISAIAH	2	2		2	6	2	3		2	2	3		3	12
19	NYIRENDA	ZAITHWA	2	3		2	5	2	4		2	2	2		2	12
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22	CHIMOMBO	MAFUNO	2	4		2	5	4	5		2	2	2		3	13
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25	KAYINGA	NATHAN	4	3		2	5	3	4		2	2	2		2	13
26	KAZIKA	JOSEPH CHITAMBALA	3	3	2	2	4		3		2	2	2		4	13
27	MACHESO	STEVEN	2	3	5	2	6		2		2	2	3		4	13
28	MALALU	CHANCY	3	3	2	2	6		3		2	2	2		3	13
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31	MAWAYA	VINCENT	3	3	3	2	5		3		2		2	2	2	13

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l when the half reactions listed above are combined into a balanced chemical equation the ratio of the coefficients $\text{Cu} : \text{MnO}_4^-$
4 : 3 is

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stoichiometric coefficient of HSO_4^- if the above equation is balanced in an acidic medium stuck use a hint report a problem
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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web aug 29 2023 some points to remember when balancing redox reactions the equation is separated into two half equations one for oxidation and one for reduction the equation is balanced by adjusting coefficients and adding H^+ and H_2O in this order balance the atoms in the equation apart from o and h

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{HNO}_2 + 2\text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} \rightarrow 2\text{HNO} + 2\text{C}_6\text{H}_8\text{O}_6 + 2\text{H}_2\text{O}$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and HNO_2 a $\text{C}_6\text{H}_8\text{O}_6$ and HNO_2 b $\text{C}_6\text{H}_8\text{O}_6$ and HNO_2 c $\text{C}_6\text{H}_8\text{O}_6$ and HNO_2 d $\text{C}_6\text{H}_8\text{O}_6$ and HNO_2

