

**TVS MATRICULATION HIGHER SECONDARY SCHOOL
TVS NAGAR, MADURAI-03.
NAS MODEL EXAM QUESTION PAPER – 2018**

ENGLISH

1. Read the passage given below and answer the questions that follow:

India has deployed aerial drones over Kaziranga National Park in Assam state in a bid to protect endangered one-horned rhinos from poachers. Kaziranga chief NK Vasu said the maiden drone flight on Monday was a “milestone in wildlife protection”. The park is home to two-thirds of the world’s one-horned rhino population and also has a large number of elephants, tigers and other wildlife.

In recent months, rhinos have been killed in large numbers by poachers. Drones and other successful anti-poaching measures have also been used by the WWF in nearby Chitwan National Park in Nepal, where the hunting of one-horned rhinos has been drastically reduced. “The Unmanned Aerial Vehicle (UAV) was up in the sky for 15 minutes. It landed safely,” Mr. Vasu said. “We hope this technology will go a long way in effective surveillance of the park.” Assam Forest Minister Rockybul Hussain said this was the first time that drones had been used for wildlife protection anywhere in India. “The UAVs will deter poachers who will now have to reckon with surveillance from air as well as on ground,” Mr. Hussain said. The minister said it would now be possible to keep an eye on the remotest parts of the 480 sq km (185 sq miles) park. The drones can fly a pre-programmed route at a maximum elevation of 200m (656ft) for up to 90 minutes, officials say.

They are light enough to be launched by hand and will be able to take images of the ground below with a still or video camera, they add. Twenty-two one-horned rhinos were killed in Kaziranga by poachers last year. In the first three years of this month, 16 rhinos have been killed, triggering a furore in Assam where the animal is seen as a symbol of regional pride and is also valued for drawing tourists to the state. A rhino census in Kaziranga last month said that there were about 2,300 of the animals in Kaziranga. BBC NEWS

Read the passage and answer the following questions:

- a) This passage states that drones
And not only for wars between nations.
- b) The drones will be able to do two things: they are
1).....
2).....
- c) 22 one horned rhinos killed by poachers means.....
- d) The introduction of drones for surveillance is a -----measure.
(one word)
- e) The author states that there has been a public outburst because 16 rhinos were killed-the term he uses is-----
- f) This passage suggests that the greatest enemy of wildlife is -----
- g) The passage begins with the idea that drones will be a “milestone in wildlife protection”. This actually means that-----

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab

questions write the balanced equation

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated CuSO_4 solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

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