

Reflective Essay

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My belief about myself as a learner when I entered the class was a positive one. I attended a college prep school for high school and went to Oregon State University last year. I had plenty of practice with different learning techniques and found what works well for me. I believe I'm more of a visual learner than an auditory learner. It is easier to see a powerpoint and retain the knowledge then listen to a lecture and have the same retention. Entering the class I thought I liked working in groups more than working individually on projects, but that has since changed. I think it is more beneficial to me to work alone because I can get more done. When I entered the class I believed that reading the material closer to the due date was a better idea because I may not remember what I had read if I didn't.

What I have discovered about myself, as a learner over the term is vast. I learned that I can read the material way in advance and just briefly skim over the text and remember just about everything, were as I thought reading closer was more beneficial. I also learned that working in groups really isn't my specialty. I prefer working by myself because I'm able to get a lot more done and not have to worry about other people forgetting their part of the project. I don't like having to worry about other people not following through. I also learned that looking at the daily record in advanced helped me

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M Woodhall



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interaction that is responsible for forming the following solutions a kno 32 in water b br in benzene c 6 h 6 c glycerol ch 22

oh ch oh ch oh in water d hcl in acetonitrile ch 3 cn hcl does not form ions in

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web solubility and solubility product mh 5.16.1 5.16.2 in this section we are going to consider the solubility of ionic solids in water solubility may be considered to be an equilibrium the equilibrium is between solid and ions in solution any ionic solid is 100% ionized in aqueous solution once it actually dissolves

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