SKILLS TRANSPARENCY

WORKSHEET



Using Mole Ratios

Use with Chapter 11. Section 11.2

For each of the following problems, write the balanced chemical equation that represents the reaction. Thru complete the table below by identifying the known substance, the anknown substance, and the mole ratio that you would use to solve each problem correctly.

Nort How & Noth

- 1. Copper(II) axide (CuO) decomposes into copper (Cu) and exygen (O₂) gas. What mass of copper will be produced by the decomposition of 1.25 kg CaQ®
- BCCuOm 22 Cum+ Oan Chart Oscar Aremonia (NH₂) is produced by the reaction of aitingen (N₂) and hydrogen (H₂) gases. How much ammonia will be produced if 22.8 r. H., reads with excess N.?
- 3. The reaction of sodium (Na) and water (H₂O) produces codium by free ide (NgOH) and hydrogen (H.) gas. What mass of evdrogen are is produced if 17.54 a NaOH improduced law the reaction?

Wards Hadis NobH + Hais 2Nam + 24000 =

 The combustion of seetic sold (HC₂H₂O₂) produces carbon diexide (CO₂) and water (HoOt-What mass of carbon discuste will be produced from the combustion of 25.0 a FIRE HEADING

S. 200 a of issuality sufficient (Eq.S.) was prepared by heating iron (Eq. and excess softer DCOa 1 2 1 2 C

Fe cs + Sch = Fe S = 2 Fe + 3 Sr = Fe S. (S). What mass of iron was used in the preparation?

Problem.	Chemical Formula of Spinos Substance	Chemical Formata of Enkages Substance	Mole Ratio
4	صن	Cu	2 mai Cu
			2 molCuO
2	+12	NH3	a mal M H3
3	MOH	Ha	2 mal 1400 H
4.	HC2 H3 03	COa	- 1 most 462 Haba
5	Fes53	Te	1 most fe, 53

Math Skills Transparency Worksheet Answers Chapter 7

R Barnett

Math Skills Transparency Worksheet Answers Chapter 7:

The Enigmatic Realm of **Math Skills Transparency Worksheet Answers Chapter 7**: Unleashing the Language is Inner Magic

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