

Honors Chemistry
Equilibrium Worksheet

Equilibrium: ICE box practice problems

1. In a 10.0L vessel at 1000K, 0.250 mol $\text{SO}_2(\text{g})$ and 0.200 mol $\text{O}_2(\text{g})$ react to form 0.162 mol $\text{SO}_3(\text{g})$ at equilibrium. What is the K_c at 1000K for this reaction?

Balanced equation: $\text{SO}_2(\text{g}) + \text{O}_2(\text{g}) \leftrightarrow \text{SO}_3(\text{g})$

	SO_2	+	O_2	\leftrightarrow	SO_3
I					
C					
E					

2. An aqueous solution of 0.00325M Fe^{3+} was reacted with 0.0020M SCN^- at 25°C. Calculate the value of the equilibrium constant for this reaction when the equilibrium concentration of FeSCN^{2+} was determined to be 0.000667M

	Fe^{3+}	+	SCN^-	\leftrightarrow	FeSCN^{2+}
I					
C					
E					

3. At a certain temperature, 2.0 mole NH_3 is introduced into a 2.0 L container, and the NH_3 partially dissociates into by the following reaction:



At equilibrium, 1.0 mol NH_3 remains, what is the value of K for this reaction?

	$2\text{NH}_3(\text{g})$	\leftrightarrow	$\text{N}_2(\text{g})$	+	$3\text{H}_2(\text{g})$
I					
C					
E					

Ice Box Practice Problems

Wenbin Ji



Ice Box Practice Problems:

ICE Handbook of Urban Drainage Practice Richard Ashley, Brian Smith, Paul Shaffer, Issy Caffoor, 2024-02-16 Written by leading experts ICE Handbook of Urban Drainage Practice provides an overview of key challenges opportunities and future directions of urban drainage in a practical accessible way An invaluable tool for local authority engineers environmental engineers drainage design operation engineers and consultants or contractors **Ice and Refrigeration**, 1926 Self-study Course 3011-G, 1991 **Essential Values-Based Practice** K. W. M. Fulford, Ed Peile, Heidi Carroll, 2012-03-01 This book will help clinicians acquire and develop the processes and skills of values based practice The aim of most patient clinician consultations is to improve health outcomes Often they succeed and patients are satisfied and empowered However some consultations are unsatisfactory and result in failure to improve health outcomes and dissatisfaction on the part of patients carers or clinicians When consultations fail to achieve the desired results the cause is not usually a failure of evidence based practice Today s clinicians are trained in evidence based medicine educated updated and appraised The most likely reason why things go wrong is a failure of values based practice not ascertaining the relevant values perspectives and acting on them in a coherent and purposeful manner If you rehearse and practise the elements of values based practice detailed in this book you will find your consultations more personally rewarding and your patients are likely to derive more benefit *Refrigerating World*, 1918 *Industrial Refrigeration*, 1907 **International Record of Medicine and General Practice Clinics** Edward Swift Dunster, James Bradbridge Hunter, Frank Pierce Foster, Charles Eucharist de Medicis Sajous, Gregory Stragnell, Henry J. Klaunberg, Félix Martí-Ibáñez, 1912 **ICE Handbook of Ground Investigation** Peter Reading, Miles Martin, 2025-07-09 ICE Handbook of Ground Investigation is a comprehensive practical guide presented in two volumes that covers the range of techniques used in conducting thorough and effective ground investigations Refrigerating World Incorporating Cold Storage & Ice Trade Journal, 1916 **Refrigerating World**, 1932 **Industrial Refrigeration**, 1958 **Popular Mechanics**, 1987-10 Popular Mechanics inspires instructs and influences readers to help them master the modern world Whether it s practical DIY home improvement tips gadgets and digital technology information on the newest cars or the latest breakthroughs in science PM is the ultimate guide to our high tech lifestyle *Merchant Plumber and Fitter*, 1916 *The Practice of Chemistry* Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can t do chemistry if they can t do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book s unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors This special PowerPoint tour of the text was created by Don Wink http://www.bfwpub.com/pdfs/wink/POC/PowerPoint_Final.ppt 832KB **Cold Storage and Ice**

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