

Name _____
stoichiometry

Test next class over

1. In the formation of carbon dioxide from carbon monoxide and oxygen, how many moles of carbon monoxide are needed to react completely with 7.0 moles of oxygen gas?
2. How many moles of oxygen are required to burn 45.8 grams of ethane gas, C_2H_6 at standard conditions?
3. How many grams of oxygen are produced by the decomposition of 55.0 grams of potassium chlorate, $KClO_3$?
4. A chemist performs the synthesis of sodium chloride from its elements. The chemist begins with 46.55 grams of sodium. How many grams of chlorine are needed?
5. How many grams of water can be prepared from 5.68 moles of hydrogen at standard conditions?
6. Suppose that an excess of propane, C_3H_8 burns in 62.0 g of O_2 . How many moles of H_2O will be formed?

Grade 1 stoichiometry Practice Test

ML Morrison



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