

CHEMISTRY

Paper & Solution

Code: A

Max. Marks: 100

Time: 3 Hrs.

Q 1. Write the formulae of any two oxoacids of phosphorus.**Answer:**(i) H_3PO_3 (Phosphorous acid)(ii) H_3PO_4 (Phosphoric acid)**Q 2.** Which would undergo S_N2 reaction faster in the following pair :**Answer:** $C_6H_5-CH_2-CH_2-Br$ (due to less steric hindrance)**Q 3.** Out of $AlCl_3$ and $NaCl$, which is more effective in causing coagulation of a negative sol and why ?**Answer:** $AlCl_3$ is more effective due to high Positive Charge density on Al^{+3} as compared to Na^+ **Q 4.** Write the formula of a compound in which the element Y forms ccp lattice and atoms of X occupy $1/3^{rd}$ of tetrahedral voids.**Answer:**

The no. of atoms Y in CCP = N no. of tetrahedral voids = 2N

$$\text{no. of X atoms} = 2N \times \frac{1}{3}$$

$$= \frac{2}{3}N$$

$$\text{Formula} \quad x \quad : \quad y$$

$$\frac{2}{3}N \quad : \quad N$$

$$= \quad 2 \quad : \quad 3$$

$$= x_2y_3$$

Grade 12 Chemistry 2015 March

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