# QUESTION 7 (Start on a new page.) The equation for the combustion of butane gas is given below.

butane(g) + 
$$13O_2(g) \rightarrow 8CO_2(g) + 10H_2O(g)$$
  $\Delta H < 0$ 

7.1 Define the term activation energy.

7.2

- Is the combustion reaction of butane exothermic or endothermic? Give a reason for the answer.
- 7.3 Draw a sketch graph of potential energy versus course of reaction for the reaction above.

Clearly indicate the following on the graph:

- Activation energy
- Heat of reaction (ΔH)
- Reactants and products

CHEMISTRY

part 3

(3)

7.4 Determine the empirical formula of butane are if it consists of 93.76% early

# **Grade 11physical Science Chemistry Part 2014 Paper 2**

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