Practice QUIZ 2

| Naturally occurring copper exists in two soctopic forms: ⁶⁵Cu and ⁶⁵Cu. The atomic mast of copper is 63.53 arms. What is the approximate natural abundance of ⁶⁵Cu? A) 67% B) 50% C) 70% D) 50% E) 30% |
|--|
| Andrew B |
| For each of the following compounds, state whether it is ionic or covalent. If it is ionic, write the symbols for the ions involved. |
| 200 08H6500s |
| Answer: (a) covalent, (b) ionic, Ba^{2n} , O^{2n} , (c) ionic, BB_{n}^{n} , OO_{n}^{2n} ; |
| For each of the following point of ions, write the symbol for the formula of the compound they will form: (a) Ca²⁺, Si²⁺ (b) NH_s²⁺, SO_s²⁺ (c) AP²⁺, Be²⁺ |
| Anner: a) Cat. (6) 0/95/5/904, (c) Allers. |
| 4. Which one of the following molecular formular is also an empirical formula? A) Celligible B) Celligible C) R2O2 D) R2P4O6 E) Celligible A some of the following molecular formular is also an empirical formula? A proper B |

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Lauren Gardner

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