## Types of Reactions Worksheet THEN Balancing!

First, begin by telling which type of rounties is taking place. Then go hack and halonce the following systems: To practice balancing, you may use the Phet Lab online. When finished, check your answers.

NaBe + \_\_\_\_ H,PO<sub>4</sub> → \_\_\_\_ Na,PO<sub>4</sub> + \_\_\_\_ HBe

Type of reaction:

C<sub>4</sub>(OH)<sub>3</sub> + M<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> → C<sub>4</sub>SO<sub>4</sub> + M(OH)<sub>3</sub>

Type of reaction:

3) Mg + \_\_\_ Fe<sub>2</sub>O<sub>2</sub> → \_\_\_ Fe + \_\_\_ MgO

Type of reaction:

4) \_\_\_\_C<sub>2</sub>H<sub>4</sub>+\_\_\_O<sub>2</sub>→ \_\_\_CO<sub>2</sub>+\_\_\_H<sub>2</sub>O

Type of reaction:

5) \_\_\_\_PISO, → \_\_\_PISO, +\_\_\_O,

Type of reaction:

6) NH, + \_\_\_ l<sub>x</sub> → \_\_\_ N, l<sub>x</sub> + \_\_\_ H,

Type of seaction:

7) \_\_\_\_H,O+\_\_\_SO, → \_\_\_\_H,SO<sub>4</sub>

Type of reaction:

8) \_\_\_\_H<sub>2</sub>SO<sub>4</sub> + \_\_\_\_NH<sub>4</sub>OH → \_\_\_\_H<sub>2</sub>O + \_\_\_\_(NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>Type of reaction: \_\_\_\_\_\_

#### **Balancing Equations Practice Worksheet**

Behavior the following agreetions:

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**Karin Nielsen-Saines** 

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