- a electron a sarrier.
- b. hard-energy wagar
- c. princ/somm term.
 - d payer-end

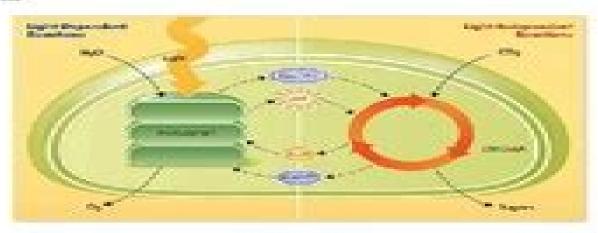


Figure 0-3

- 22. Which chemical shown in Figure 6-3 m an electron carmer molecule?
 - A 25.40
 - b. carbon dicadds
 - BRADET
 - d. coorden
- 23. What makes certain molecules good electron carriers?
 - They can accept electrons and transfer most of their energy to another molecule.
 - They are very large molecules, so they have lots of room to carry many electrons.
 - The can absorb sunlight, which is where all of the high-energy electrons come from
 - d. They are carboly-frates and have a lot of energy which allows them to carry electrons.
- 34. Why are electron-carriers needed for transporting electrons from one part of the chloroplast to another?
 - a. High-energy electrons would be destroyed.
 - b. High-energy electrons are highly reactive.
 - c. High-energy electrons are not soluble in cytoplasm.
 - d. High-energy electrosa get their energy from electron carriers.
- 26. A student is collecting the gas given off from a plant in bright vanishis at a temperature of 27°C. The cas being collected is probably.

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