

Manometer Example



$$\begin{aligned} p_{\text{gas}} &= p_{\text{atm}} \pm \rho g h \\ &= 101 \text{ kPa} - (13,550 \frac{\text{kg}}{\text{m}^3}) (9.81 \frac{\text{m}}{\text{s}^2}) (0.3 \text{ m}) \\ &= 101 \text{ kPa} - 19.9 \text{ kPa} \end{aligned}$$

$$p_{\text{gas}} = 81.1 \text{ kPa}$$

Manometer Practice Problems

Marcel A. Müller



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