ADVANCED HONORS CHEMISTRY - CHAPTER 13 NAME: STATES OF MATTER DATE: MANOMETER PROBLEMS WKST - ANSWERS - V5 PAGE:

- 1. In the diagram below h is 15 mm Hg.
 - A. What is the pressure of the gas sample in mm Hg?
 - B. What is the pressure of the gas sample in kPa?

DATA TABLE		
h=15 mm Hg	P _{Gan Sample} = ?	

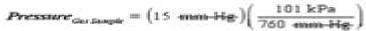
This is a close-ended manometer problem. .. The height of the Hg is the gas pressure.

Pressure Gas Sample = Height of the Hg



$$P_{Gas Sample} = 15 \text{ mm Hg}$$

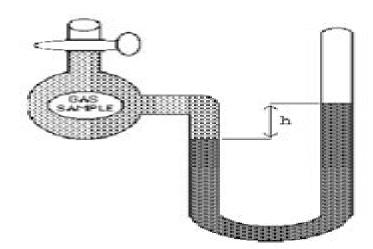




$$\frac{760 \text{ mm Hg}}{101 \text{ kPa}} = \frac{15 \text{ mm Hg}}{x}$$

$$\frac{760 \text{ mm-Hig}}{15 \text{ mm-Hig}} = \frac{101 \text{ kPa}}{x}$$

$$P_{Gas\ Sample} = 2.0 \text{ kPa}$$



Manometer Practice Problems With Answers

Christopher R. Hren, John T. Moore, Peter J. Mikulecky

Manometer Practice Problems With Answers:

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