

Graphing Periodic Trends Activity

Name: _____ Period: _____

Use the information in these tables to complete the graph as described below

Symbol	Atomic Radius (Picometers)	First Ionization energy (kJoules/mole)	Electronegativity (4-point scale)
H	31	1312	2.1
Li	128	520	1.0
Na	166	496	0.9
K	203	410	0.8
Rb	220	403	0.8
Cs	244	376	0.7

Symbol	Atomic Radius (Picometers)	First Ionization energy (kJoules/mole)	Electronegativity (4-point scale)
Na	166	496	0.9
Mg	141	738	1.2
Al	121	578	1.5
Si	111	787	1.8
P	107	1012	2.1
S	105	1000	2.5
Cl	102	1251	3.0
Ar	106	1521	---

1. Using colored pencils or pen, list under the symbol (in this order!) the Atomic Radius, First Ionization Energy, and Electronegativity.
2. Use a different color for each property. Example: Write all of the Atomic radius values in red, all of the First ionization energies in green, and all of the Electronegativities in blue. Of course, you can pick any colors that are available, as long as you are consistent.
3. Observe the trends in each property as you go down the Alkali metal group, and as you go across Period 3.
4. Write out each of the statements written on the opposite side, completing each statement with the observed trend (increase or decrease)

H	<p>Across a period (→) atomic radius tends to _____.</p> <p>Across a period (→) first ionization energy tends to _____.</p> <p>Across a period (→) electronegativity tends to _____.</p>						
Li							
Na 166 496 0.9	Mg	Al	Si	P	S	Cl	Ar
K	<p>Down a Group (↓) atomic radius tends to _____.</p> <p>Down a Group (↓) first ionization energy tends to _____.</p> <p>Down a Group (↓) electronegativity tends to _____.</p>						
Rb							
Cs							

Graphing Lab Periodic Trends Answer Key

Jo A. Beran



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The Science Teachers Bulletin ,1989 *Laboratory Manual* Jo A. Beran,1990 **Prentice Hall Science Explorer**
Michael J. Padilla,2002 *Prentice Hall Chemistry* ,2000 **Microscale General Chemistry Laboratory** Zvi
Szafran,Ronald M. Pike,Judith C. Foster,2012-04-13 In the past two decades microscale techniques have soared in popularity
because these techniques minimize exposure to potentially dangerous chemicals in the lab drastically cut the amount of
chemical waste lower costs and reduce risks of chemical fires and explosions The result is a safer and healthier laboratory
environment Now with Microscale General Chemistry Laboratory with Selected Macroscale Experiments Second Edition you
can bring these techniques into your own chemistry lab Thoroughly revised with updated experiments the new Second
Edition continues to offer a large variety of well designed easy to follow experiments as well as thorough background
information and an outstanding selection of questions and problems **Metaheuristics for Vehicle Routing Problems**
Nacima Labadie,Christian Prins,Caroline Prodhon,2016-02-23 This book is dedicated to metaheuristics as applied to vehicle
routing problems Several implementations are given as illustrative examples along with applications to several typical vehicle
routing problems As a first step a general presentation intends to make the reader more familiar with the related field of
logistics and combinatorial optimization This preamble is completed with a description of significant heuristic methods
classically used to provide feasible solutions quickly and local improvement moves widely used to search for enhanced
solutions The overview of these fundamentals allows appreciating the core of the work devoted to an analysis of
metaheuristic methods for vehicle routing problems Those methods are exposed according to their feature of working either
on a sequence of single solutions or on a set of solutions or even by hybridizing metaheuristic approaches with others kind of
methods **The World of Chemistry** Jones,Jones Staff,1996 Scientific and Technical Aerospace Reports ,1986
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Table of Contents Graphing Lab Periodic Trends Answer Key

1. Understanding the eBook Graphing Lab Periodic Trends Answer Key
 - The Rise of Digital Reading Graphing Lab Periodic Trends Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Graphing Lab Periodic Trends Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Graphing Lab Periodic Trends Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Graphing Lab Periodic Trends Answer Key
 - Personalized Recommendations

- Graphing Lab Periodic Trends Answer Key User Reviews and Ratings
- Graphing Lab Periodic Trends Answer Key and Bestseller Lists
- 5. Accessing Graphing Lab Periodic Trends Answer Key Free and Paid eBooks
 - Graphing Lab Periodic Trends Answer Key Public Domain eBooks
 - Graphing Lab Periodic Trends Answer Key eBook Subscription Services
 - Graphing Lab Periodic Trends Answer Key Budget-Friendly Options
- 6. Navigating Graphing Lab Periodic Trends Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Graphing Lab Periodic Trends Answer Key Compatibility with Devices
 - Graphing Lab Periodic Trends Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Graphing Lab Periodic Trends Answer Key
 - Highlighting and Note-Taking Graphing Lab Periodic Trends Answer Key
 - Interactive Elements Graphing Lab Periodic Trends Answer Key
- 8. Staying Engaged with Graphing Lab Periodic Trends Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Graphing Lab Periodic Trends Answer Key
- 9. Balancing eBooks and Physical Books Graphing Lab Periodic Trends Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Graphing Lab Periodic Trends Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Graphing Lab Periodic Trends Answer Key
 - Setting Reading Goals Graphing Lab Periodic Trends Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Graphing Lab Periodic Trends Answer Key
 - Fact-Checking eBook Content of Graphing Lab Periodic Trends Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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