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# Making Practice Fun 17 Answers

**J Elliott**



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web allow the gaseous phase thermodynamics to be curve fitted to 300k about 142k lower than the normal boiling point 542k 2 of

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web as a general trend vapor pressures of liquids at ambient temperatures increase with decreasing boiling points this is illustrated in the vapor pressure chart see right that shows graphs of the vapor pressures versus temperatures for a variety of liquids

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web sep 21 2022 a vapor pressure curve is a graph of vapor pressure as a function of temperature to find the normal boiling point of liquid a horizontal line is drawn from the y axis at a pressure equal to standard pressure

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web the graph of the vapor pressure of water versus temperature in figure page index 3 indicates that the vapor pressure of water is 68 kpa at about 90 c thus at about 90 c the vapor pressure of water will equal the atmospheric pressure in

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