

042723 PHYSICAL SCIENCE CH 17 TEST REVIEW

DAVIS

NAME: _____

04/27/23

1. What are the 4 general characteristics of metals?

Malleable, ductile, shiny, good conductors

2. Define ductile.

able to be drawn out into a thin wire.

3. Define malleable.

able to be hammered or pressed permanently out of shape without breaking or cracking.

4. What action do metals prefer to take with their valence electrons during bonding w/ nonmetals?

Give them away

5. What happens to their charge when they do so?

Stays positive

6. What do we call atoms with this charge?

Cations

7. What do we call the type of bonding that occurs when metals and nonmetals bond?

Ionic bond

8. What type of bond forms between metals and other metallic atom?

Metallic bond

9. The electrons in metallic bonds exist in a "___cloud___" on the surface.

10. Which important characteristic of metals does this permit?

luster

11. What group contains the Alkali Metals?

IA

12. What is the name of group IIA?

Alkali Earth Metals

Holt Physical Science Chapter17 Sectionreview Answers

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Table of Contents Holt Physical Science Chapter17 Sectionreview Answers

1. Understanding the eBook Holt Physical Science Chapter17 Sectionreview Answers
 - The Rise of Digital Reading Holt Physical Science Chapter17 Sectionreview Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Physical Science Chapter17 Sectionreview Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Physical Science Chapter17 Sectionreview Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Physical Science Chapter17 Sectionreview Answers

- Personalized Recommendations
 - Holt Physical Science Chapter17 Sectionreview Answers User Reviews and Ratings
 - Holt Physical Science Chapter17 Sectionreview Answers and Bestseller Lists
5. Accessing Holt Physical Science Chapter17 Sectionreview Answers Free and Paid eBooks
 - Holt Physical Science Chapter17 Sectionreview Answers Public Domain eBooks
 - Holt Physical Science Chapter17 Sectionreview Answers eBook Subscription Services
 - Holt Physical Science Chapter17 Sectionreview Answers Budget-Friendly Options
 6. Navigating Holt Physical Science Chapter17 Sectionreview Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Physical Science Chapter17 Sectionreview Answers Compatibility with Devices
 - Holt Physical Science Chapter17 Sectionreview Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Physical Science Chapter17 Sectionreview Answers
 - Highlighting and Note-Taking Holt Physical Science Chapter17 Sectionreview Answers
 - Interactive Elements Holt Physical Science Chapter17 Sectionreview Answers
 8. Staying Engaged with Holt Physical Science Chapter17 Sectionreview Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Physical Science Chapter17 Sectionreview Answers
 9. Balancing eBooks and Physical Books Holt Physical Science Chapter17 Sectionreview Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Physical Science Chapter17 Sectionreview Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Holt Physical Science Chapter17 Sectionreview Answers
 - Setting Reading Goals Holt Physical Science Chapter17 Sectionreview Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Holt Physical Science Chapter17 Sectionreview Answers

- Fact-Checking eBook Content of Holt Physical Science Chapter17 Sectionreview Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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