#### CHEMICAL EQUILIBRIUM MULTIPLE CHOICE QUESTIONS

Read the questions properly before answering.

- It is a condition that occurs when opposing reactions are proceeding at equal rates.
  - a. Chemical kinetics
  - b. Chemical equilibrium
  - c. Chemical reaction
  - d. Chemical equation
- 2. Which of the following is not included when writing equilibrium constant expression?
  - a. gases
  - b. aqueous solutions
  - c. liquids
  - d. pure liquids and solids
- 3. Equilibrium constant greater than 1 means
  - a. equilibrium constant lies to the right, products predominate.
  - b. equilibrium constant lies to the left, reactants predominate.
  - c. equilibrium constant lies to the right, reactants predominate.
  - d. equilibrium constant lies to the left, products predominate.
- What does it mean when the Keq < 1?</li>
  - Equilibrium constant lies to the right, products predominate.
  - b. Equilibrium constant lies to the left, reactants predominate.
  - c. Equilibrium constant lies to the right, reactants predominate.
  - Equilibrium constant lies to the left, products predominate.
- What is the equilibrium constant expression of aA + bB ≠ cC + dD?

a. 
$$Keq = \frac{\{a\}^{d} \{b\}^{B}}{\{e\}^{C} \{a\}^{D}}$$

c. 
$$Keq = \frac{\{C\}^d \{D\}^d}{\{a\}^d \{b\}^B}$$

b. 
$$Keq = \frac{[A]^a [B]^b}{[C]^a [D]^d}$$

d. 
$$Keq = \frac{\{CP^{\dagger}[D]^d}{\{A\}^d \{B\}^b}$$

- 6. Which of the following is not a factor that affects chemical equilibrium?
  - a. concentration
  - b. temperature
  - c. pressure and volume
  - d. density
- 7. Equilibria in which all substances are in the same phases are called
  - a. homogeneous equilibria.
  - b. heterogeneous equilibria.
  - equilibrium constant
  - d. equilibrium constant expression.
- 8. Equilibria in which two or more phases are present in the reaction are called
  - a. homogeneous equilibria.
  - b. heterogeneous equilibria.
  - c. equilibrium constant
  - d. equilibrium constant expression.

# **Multiple Choice Concept Questions On Equilibrium**

Dr. Anupam Agrawal, Mrs. Sharad Agrawal

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