

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পদ্ধতির প্রয়োগের

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

সংকেত, $m = 20 \text{ g}$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = 1$

উদাহরণে দ্রবিত X-পদার্থিতে 10 mg এর মধ্যে কয় পরিমাণ দ্রব
যে X-পদার্থিতে 0.5 g 2% $\left(\frac{W}{V}\right)$ HCl দ্রব দিতে কাজ করা হয়েছে।

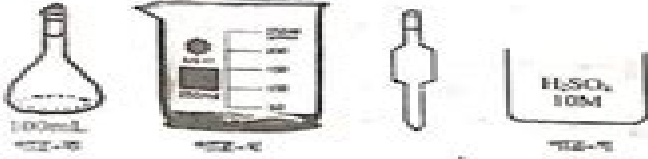
যদিও X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও
Y-পদার্থটি হলো ফার্মা আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির
মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা
হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
নিয়ন্ত্রিত বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05
থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে।
এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2S এর পরিবর্তে
কোয়েলিনাইট ব্যবহার করা হয় ফলে পরিবেশ দূষণ ঘটে না।



অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

চিত্র-১০



১. টাইটার দ্রবক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঞ্জার প্রকৃতিটি অঙ্কন করে এবং
ব্যাখ্যা করে।
৪. উদাহরণের অংশের দ্রবণ থেকে উপযুক্ত কাচ পাত্র ব্যবহার
করে 1M ঘনমাত্রার দ্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪
ও না প্রদেয় উত্তর

৫. বিশ্লেষণের নিমিত্ত বীমের উপর টাইটার স্থাপন করলে বীমের প্রতি
দিকের স্কেল যে দর পাওয়া যায়, তাকে টাইটার দ্রবক বলে।

৬. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
নিয়ন্ত্রিত বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ
(0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশ দূষণ উল্লেখযোগ্যভাবে কমে আসে।
অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশ দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশ বান্দব।

৭. উদাহরণের দ্রবণটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M
দ্রবণ। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M
দ্রবণ যখন পাত্রে H_2SO_4 দ্রবণ। পাত্রে H_2SO_4 পদার্থকে ক্যাথোডী হ্যাঞ্জার
একভাবে অঙ্কন করা হয়। পাত্রে H_2SO_4 এর অন্য ক্ষয়কারী হ্যাঞ্জার
একটি নিয়ে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রকৃতি	ক্ষয়কারী (Corrosive)
আয়তন	ক্ষয়কারী রাসায়নিক উপাদান কয়েক সে মিলিমিটার রাসায়নিক দ্রব্যকে খুঁজার দর সংস্পর্শে তুলে ও অন্যদিকে পাত্রে রাখলে ক্ষয়কারী ক্ষতি সাধিত হয়। এতে তুলে পুড়ে যায় এবং অন্যদিকে পাত্রে পড়ে যায়।
উদাহরণ	পাত্রে কঠিন এসিডসমূহ (HCl , HNO_3 , H_2SO_4), পাত্রে দ্রব (NaOH , KOH), H_2O_2 , অক্সিজেন লিকার প্রভৃতি দ্রবণ ইত্যাদি।
সংরক্ষণ পদ্ধতি	- পাত্রে এসিডকে পোতলে রাখতে হবে। - পাত্রে ক্ষয়কারী, জারক বা ক্ষয়কারী পদার্থকে পোতলে রাখতে হবে।

৮. উদাহরণের দ্রবণটি হলো 10M H_2SO_4 দ্রবণ এবং এর থেকে 1M
ঘনমাত্রার দ্রবণ প্রস্তুত করতে হবে 1ম ও ২য় চলকের ব্যবহার করতে
হবে।

1ম চলকের দ্রবণটি হলো 100 ml. অক্সিজেনমিত্রিক দ্রবণ ও এর দি হলো
নিম্নের। যেহেতু অক্সিজেনমিত্রিক দ্রবণের আয়তন 100 ml. তাই
আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 দ্রবণ তৈরি করতে হবে।
দ্রবণের সন্নিবেশ দ্রব হতে আনত।

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H_2SO_4 দ্রবণের সঙ্গে পানি যোগ করতে হবে
(100-10) mL = 90 mL.
অতএব, নিম্নের দ্রবণে 10 ml দ্রবণ ঘনমাত্রার দ্রবণ নিয়ে তার
সঙ্গে অক্সিজেনমিত্রিক দ্রবণে 90 ml পানি যোগ করলে আমাদের
কামিত ঘনমাত্রার দ্রবণ পাওয়া যাবে।

চিত্র-১১



১. অক্সিজেনমিত্রিক কয়েক বসে?
২. সাদা দ্রবক K, এর ঘন কখনো অসীম হয় না-ব্যাখ্যা
করে।
৩. A দ্রবের হ্যাঞ্জারে উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C দ্রবের সাথে সম্পর্কিত রাসায়নিক দ্রবণের
ব্যবহারকালে খান্সা কৃতি দ্রবণে কবীর উপর সম্পর্কে
বিশ্লেষণ করে। ৪
ও না প্রদেয় উত্তর

৯. অক্সিজেনমিত্রিক এক ধরনের রাসায়নিক বা কৃত্রিম দ্রবণ সঞ্চেদ
বা মুদ্রত খান্সা উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে তুলে
খান্সার সঞ্চেদ করে।

১০. সামান্য K, এর ঘন কখনো অসীম বা শূন্য হতে পারে না।
উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উদাহরণী মাইক্রোপদ্ধতি
বিশ্লেষণ করি-
 $A + B \longrightarrow C + D$

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JS Bruner



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