

LIGAND FIELD THEORY

Chemists often consider the d-orbitals of a central metal ion as being unsplit. The more we learn about the splitting of these orbitals, the more we learn about the properties of the complex. This is the basis of Ligand Field Theory. Ligand field theory is a way of describing the effect of ligands on the energy of the d-orbitals of a central metal ion.

There are two main types of ligands: strong field ligands and weak field ligands. Strong field ligands are those that cause a large splitting of the d-orbitals, while weak field ligands are those that cause a small splitting. The strength of the ligand field is determined by the nature of the ligand and the nature of the metal ion. The strength of the ligand field is also determined by the geometry of the complex. The strength of the ligand field is also determined by the nature of the metal ion.

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The ligand field is

the energy of the d-orbitals of a central metal ion as a function of the nature of the ligands. The ligand field is a way of describing the effect of ligands on the energy of the d-orbitals of a central metal ion.



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Introduction To Ligand Field Theory

L Cohen



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