

Empirical and Molecular Formula Worksheet ANSWER KEY

Write the empirical formula for the following compounds.

- 1) C_6H_6 **CH**
- 6) C_8H_{14} **C_4H_7**
- 7) WO_2 **WO_2**
- 8) $C_2H_6O_2$ **CH_3O**
- 9) $X_{10}Y_{12}$ **X_5Y_6**
- 6) A compound with an empirical formula of C_2OH_4 and a molar mass of 88 grams per mole. What is the molecular formula of this compound? **$C_8O_4H_{16}$**
- 7) A compound with an empirical formula of C_4H_4O and a molar mass of 136 grams per mole. What is the molecular formula of this compound? **$C_8H_8O_2$**
- 8) A compound with an empirical formula of $CFBrO$ and a molar mass of 253.8 grams per mole. What is the molecular formula of this compound? **$C_2F_2Br_2O_2$**
- 9) A compound with an empirical formula of C_3H_3N and a molar mass of 46 grams per mole. What is the molecular formula of this compound? **$C_6H_6N_2$**
- 10) A well-known reagent in analytical chemistry, dimethylglyoxime, has the empirical formula C_2H_5NO . If its molar mass is 116.1 g/mol, what is the molecular formula of the compound? **$C_8H_{10}N_2O_2$**
12. A certain blue solid contains 36.84% N and 63.16% O. What is the empirical formula of this compound? **The ratios are $N_{1.50}O_{1.50}$. Since 1.50 is not close to a whole number, we multiply *both* subscripts by 2. The empirical formula is thus N_2O_2 . (The name is dinitrogen trioxide.)**
13. A sample of indium chloride weighing 0.5000 g is found to contain 0.2404 g of chlorine. What is the empirical formula of the indium compound? **$InCl_2$**
14. An unknown compound was found to have a percent composition as follows: 47.0 % potassium, 14.5 % carbon, and 38.5 % oxygen. What is its empirical formula? If the true molar mass of the compound is 166.22 g/mol, what is its molecular formula? **$K_2C_3O_4$**
15. Rubbing alcohol was found to contain 60.0 % carbon, 13.4 % hydrogen, and the remaining mass was due to oxygen. What is the empirical formula of rubbing alcohol? **C_3H_8O**

Empirical And Molecular Formula Answer

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CLASS 11 CHEMISTRY 5 SOLVED "CASE STUDIES" Priti Singhal,2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging

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