## Mole/Mole Ratio Problems

1. \_\_N<sub>2</sub> + \_\_H<sub>2</sub> → \_\_NH<sub>3</sub>

- a. How many moles of hydrogen are needed to completely react with two moles of nitrogen?
- b. How many moles of nitrogen trihydride can be produced with 5 moles of nitrogen?
- c. How many moles of nitrogen are needed to produce .5 moles of nitrogen trihydride?
- d. How many liters of nitrogen trihydride can be produced from 24 liters of nitrogen gas?
  - \_\_KCIO<sub>3</sub> → \_\_KCI + \_\_O<sub>2</sub>
  - a. How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate?
  - b. How many moles of potassium chloride are produced by the decomposition of .75 moles of potassium chlorate?
  - c. How many moles of potassium chlorate are needed to produce 100 moles of oxygen?
  - d. How many liters of oxygen will the decomposition of 100 liters of potassium chlorate produce?

# **Mole Ratios Worksheet Chapter 12**

R Bogdan

**Mole Ratios Worksheet Chapter 12:** 

## Mole Ratios Worksheet Chapter 12 Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "Mole Ratios Worksheet Chapter 12," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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