

Name _____ answer key _____ Date _____

Stoichiometry: Mass – Mole Worksheet

Based on the following chemical equation, please answer the following questions!



Calculate the amount of moles of NaCl needed to produce 75 g of FeCl₃.

9 FeCl ₃	1 mole FeCl ₃	mole NaCl
9 FeCl ₃	9 FeCl ₃	mole FeCl

Calculate the amount of Fe₂ (SO₄)₃ in grams produce 215 g Na₂SO₄.

9 Na ₂ SO ₄	1 mole Na ₂ SO ₄	mole Fe ₂ (SO ₄) ₃	9 Fe ₂ (SO ₄) ₃
9 Na ₂ SO ₄	9 Na ₂ SO ₄	mole Na ₂ SO ₄	1 mole Fe ₂ (SO ₄) ₃

Calculate the amount in grams of FeCl₃ produced with 95g of Fe (SO₄)₃.

9 Fe ₂ (SO ₄) ₃	1 mole Fe ₂ (SO ₄) ₃	mole FeCl ₃	9 FeCl ₃
9 Fe ₂ (SO ₄) ₃	9 Fe ₂ (SO ₄) ₃	mole Fe(SO ₄) ₃	1 mole FeCl ₃

Calculate the amount of moles of Fe₂(SO₄)₃ needed to react with 123.5 g of NaCl

9 NaCl	1 mole NaCl	mole Fe(SO ₄) ₃
9 NaCl	9 NaCl	mole NaCl

Based on the following chemical equation, please answer the following questions!



9 Ca(HCO ₃) ₂	1 mole Ca(HCO ₃) ₂	mole CaCO ₃
9 Ca(HCO ₃) ₂	9 Ca(HCO ₃) ₂	mole Ca(HCO ₃) ₂

Mole Ratio Conversion Test Answers

Albert A Gayle



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