

Q-7: State True and False. If False, state reasons.

- In a closed vessel, the reaction between heated iron and steam is reversible.
- Aluminium is highly corrosion resistant.
- Strong heating of green FeSO_4 crystals results in a FeO residue.
- A chemical reaction is always accompanied by a temperature increase.
- Respiration is the process by which glucose in the body is oxidised.

Q-8: Examples of decomposition reactions are provided below



These reactions represent respectively

- thermal decomposition, thermal decomposition
- thermal decomposition, electrolytic decomposition
- photolytic decomposition, electrolytic decomposition
- photolytic decomposition, thermal decomposition

Q-9: Which of the following substances does not produce CO_2 when exposed to dilute acid?

- Lime
- Limestone
- Baking soda
- Marble

Q-10: Match the applications in Column I with the chemical substances in Column II.

Column I	Column II
A) Antacid	1) Washing soda
B) Decolourisation	2) Sodium chloride
C) Production of H_2 and Cl_2	3) Bleaching powder
D) Preparation of glass	4) Baking soda

Q-11: Classify the following reactions as decomposition, combination, displacement, or double displacement.

- $\text{Cl}_2\text{(g)} + 2\text{NaBr(aq)} \rightarrow \text{Br}_2 + 2\text{NaCl(aq)}$
- $\text{CaO(s)} + \text{H}_2\text{O(l)} \rightarrow \text{Ca(OH)}_2\text{(aq)}$
- $\text{Ni(NO}_3)_2\text{(aq)} + 2\text{NaOH(aq)} \rightarrow \text{Ni(OH)}_2\text{(s)} + 2\text{NaNO}_3\text{(aq)}$
- $2\text{CuO(s)} \rightarrow 2\text{Cu(s)} + \text{O}_2\text{(g)}$

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Survey of Science History & Concepts Parent Lesson Plan, 2013-08-01

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