

Name _____ Class _____ Date _____

Assessment

Chapter Test A

Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
 - a slippery feel
 - the ability to conduct an electric current
 - reactivity with metals
- _____ 2. Which is an example of a weak acid?
- HCl
 - H_2SO_4
 - H_3PO_4
 - HNO_3
- _____ 3. Acetic acid is found in
- vinegar.
 - the stomach.
 - antacids.
 - oranges.
- _____ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
 - tastes bitter.
 - dissociates to form OH^- ions.
 - All of the above
- _____ 5. Which of the following is a Brønsted-Lowry base?
- HCl
 - HCO_3^-
 - H_3O^+
 - H_3PO_4
- _____ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(aq) + \text{CH}_3\text{NH}_2(aq) \rightleftharpoons \text{HC}_2\text{O}_4^-(aq) + \text{CH}_3\text{NH}_3^+(aq),$$
- which of these is a conjugate acid-base pair?
- $\text{H}_2\text{C}_2\text{O}_4(aq)$ and $\text{CH}_3\text{NH}_2(aq)$
 - $\text{H}_2\text{C}_2\text{O}_4(aq)$ and $\text{CH}_3\text{NH}_3^+(aq)$
 - $\text{CH}_3\text{NH}_2(aq)$ and $\text{CH}_3\text{NH}_3^+(aq)$
 - $\text{HC}_2\text{O}_4^-(aq)$ and $\text{CH}_3\text{NH}_3^+(aq)$

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Table of Contents Modern Chemistry Chapter 5 Sample Tests

1. Understanding the eBook Modern Chemistry Chapter 5 Sample Tests
 - The Rise of Digital Reading Modern Chemistry Chapter 5 Sample Tests
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Chapter 5 Sample Tests
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Chapter 5 Sample Tests
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Chapter 5 Sample Tests
 - Personalized Recommendations
 - Modern Chemistry Chapter 5 Sample Tests User Reviews and Ratings
 - Modern Chemistry Chapter 5 Sample Tests and Bestseller Lists
5. Accessing Modern Chemistry Chapter 5 Sample Tests Free and Paid eBooks

- Modern Chemistry Chapter 5 Sample Tests Public Domain eBooks
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 - Modern Chemistry Chapter 5 Sample Tests Budget-Friendly Options
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 - Modern Chemistry Chapter 5 Sample Tests Enhanced eBook Features
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- Adjustable Fonts and Text Sizes of Modern Chemistry Chapter 5 Sample Tests
 - Highlighting and Note-Taking Modern Chemistry Chapter 5 Sample Tests
 - Interactive Elements Modern Chemistry Chapter 5 Sample Tests
8. Staying Engaged with Modern Chemistry Chapter 5 Sample Tests
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Chapter 5 Sample Tests
9. Balancing eBooks and Physical Books Modern Chemistry Chapter 5 Sample Tests
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Chapter 5 Sample Tests
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modern Chemistry Chapter 5 Sample Tests
- Setting Reading Goals Modern Chemistry Chapter 5 Sample Tests
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Chemistry Chapter 5 Sample Tests
- Fact-Checking eBook Content of Modern Chemistry Chapter 5 Sample Tests
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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