

Measurements and Calculations

MULTIPLE CHOICE: On the line at the left of each statement, write the letter of the choice that best completes the statement or answers the question.

- 1) The four major stages of the scientific method are .
 - a) observing, formulating hypotheses, testing hypotheses, and theorizing
 - b) observing, generalizing, theorizing, and formulating hypotheses
 - c) observing, generalizing, theorizing, and testing hypotheses
 - d) collecting data, predicting, testing hypotheses, and theorizing
- 2) By making several measurements with the same balance, a chemist obtained values of 5.224 g, 5.235 g, and 5.259 g for the mass of a sample. Without knowing the actual mass of the sample, we can tell that these measurements have .
 - a) good precision
 - b) good accuracy
 - c) poor precision
 - d) poor accuracy
- 3) Which equation is an inverse proportion?
 - a) $V = (1/k)T$
 - b) $F = -kX^2$
 - c) $PV = k$
 - d) $Y = kX - 8$
- 4) The distance between the sun and Earth is 150 million km. Light travels at a speed of 3.0×10^8 m/s. Dividing the distance by the speed and multiplying by the number of meters in a kilometer will result in units of .
 - a) seconds
 - b) meters per second
 - c) kilometers squared
 - d) kilometers per second
- 5) The measurement that has only non-significant zeros is.
 - a) 506 mL
 - b) 60.0 mL
 - c) 0.0037 mL
 - d) 400. mL
- 6) If some measurements agree closely but differ widely from the actual value, these measurements are .
 - a) both accurate and precise
 - b) neither precise nor accurate
 - c) precise but not accurate
 - d) accurate but not precise
- 7) If two quantities are directly proportional, when one quantity increases by 10 percent, the other .
 - a) increases by 10 percent
 - b) decreases by 10 percent
 - c) stays the same
 - d) may increase at a different rate
- 8) In division and multiplication, the answer must not have more significant figures than the .
 - a) number in the calculation with the fewest significant figures
 - b) number in the calculation with the most significant figures
 - c) average number of significant figures in the problem
 - d) total number of significant figures in the problem

Modern Chemistry Chapter 2 Test Answers

**David W. Oxtoby, H. Pat Gillis, Laurie J.
Butler**



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