

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

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Solutions Manual for Perspectives on Structure and Mechanism in Organic Chemistry Felix A.

Carroll,2023-04-11 SOLUTIONS MANUAL FOR PERSPECTIVES ON STRUCTURE AND MECHANISM IN ORGANIC CHEMISTRY Based on the author s first hand classroom experience this solutions manual complements the 3rd edition of Perspectives on Structure and Mechanism in Organic Chemistry The solutions to the 438 textbook problems help students increase their understanding of physical organic chemistry and more than 550 references stimulate their engagement with the chemical literature **Modern Chemistry** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 **Modern Biophysical Chemistry** Peter Jomo Walla,2014-09-22 This updated and up to date version of the first edition continues with the really interesting stuff to spice up a standard biophysics and biophysical chemistry course All relevant methods used in current cutting edge research including such recent developments as super resolution microscopy and next generation DNA sequencing techniques as well as industrial applications are explained The text has been developed from a graduate course taught by the author for several years and by presenting a mix of basic theory and real life examples he closes the gap between theory and experiment The first part on basic biophysical chemistry surveys fundamental and spectroscopic techniques as well as biomolecular properties that represent the modern standard and are also the basis for the more sophisticated technologies discussed later in the book The second part covers the latest bioanalytical techniques such as the mentioned super resolution and next generation sequencing methods confocal fluorescence microscopy light sheet microscopy two photon microscopy and ultrafast spectroscopy single molecule optical electrical and force measurements fluorescence correlation spectroscopy optical tweezers quantum dots and DNA origami techniques Both the text and illustrations have been prepared in a clear and accessible style with extended and updated exercises and their solutions accompanying each chapter Readers with a basic understanding of biochemistry and or biophysics will quickly gain an overview of cutting edge technology for the biophysical analysis of proteins nucleic acids and other biomolecules and their interactions Equally any student contemplating a career in the chemical pharmaceutical or bio industry will greatly benefit from the technological knowledge presented Questions of differing complexity testing the reader s understanding can be found at the end of each chapter with clearly described solutions available on the Wiley VCH textbook homepage under www.wiley-vch.de/textbooks **Modern Chemistry** Raymond E. Davis,1999 2000 2005 State Textbook Adoption Rowan Salisbury **Science For Tenth Class Part 2 Chemistry** Lakhmir Singh & Manjit Kaur, A series of six books for Classes IX and X according to the CBSE syllabus Each class divided into 3 parts Part 1 Physics Part 2 Chemistry Part 3 Biology **Science For Tenth Class Part 2 Chemistry** LAKHMIR SINGH, A series of six books for Classes IX and X according to the CBSE syllabus **Oswaal CTET (Central Teachers Eligibility Test) 15 Previous Years Solved Papers (2013 - 2023) Paper - II (Classes 6 to 8) (Social Science & Social Studies) Yearwise For 2024 Exam** Oswaal Editorial Board,2023-11-04

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