

## Test Quantities in Chemical Reactions SCH3U #13

Name: \_\_\_\_\_

\_\_\_K \_\_\_T \_\_\_C \_\_\_A Assessment of Learning (Product)  
 10 20 8 7

Multiple Choice: Circle the most correct answer. (10K)

1. What is the molar mass of  $\text{AuCl}_3$ ?

- A) 96 g B) 130 g C) 232.5 g D) 303.6 g

2. Determine the mass in grams of 0.0489 mol cobalt.

- A) 0.00812 g B) 0.294 g C) 2.88 g D) 0.000830 g

3. The number of molecules of hydrogen peroxide found in 3.15 mol  $\text{H}_2\text{O}_2$  is:

- A) 3 B)  $1.90 \times 10^{23}$  C)  $3.80 \times 10^{23}$  D)  $7.60 \times 10^{23}$

4. The percentage of oxygen, by mass, in  $\text{Mg}_3(\text{PO}_4)_2$  is:

- A) 15.9% B) 23.6% C) 27.7%  
 D) 48.7% E) None of the above

5. An unknown gas is found to consist of 10.48 g of nitrogen and 11.96 g of oxygen. What is the percentage composition of this gas?

- A) 28.6% N, 71.4 % O B) 46.7%N, 53.3% O  
 C) 53.3% N, 46.7% O D) 74.5% N, 25.5% O

6. A compound contains 38.7% K, 13.9% N, and 47.4% O by mass. What is the empirical formula of the compound?

- A)  $\text{KNO}_3$  B)  $\text{K}_2\text{N}_2\text{O}_3$  C)  $\text{KNO}_2$  D)  $\text{K}_2\text{NO}_3$  E)  $\text{K}_4\text{NO}_3$

7. If the molar mass of a hydrocarbon is 26.0 g/mol, and its empirical formula is  $\text{CH}$ , its molecular formula is

- A)  $\text{CH}$  B)  $\text{C}_2\text{H}_2$  C)  $\text{C}_3\text{H}_3$  D)  $\text{C}_4\text{H}_4$  E)  $\text{C}_5\text{H}_5$

8. Consider the following reaction:  $\text{Ca}(\text{OH})_{2(s)} + \text{Al}(\text{SO}_4)_{3(s)} \rightarrow \text{CaSO}_{4(s)} + \text{Al}(\text{OH})_{3(s)}$

When balanced, the mole ratio for calcium hydroxide reacting with aluminum sulphate is

- A) 1:1 B) 2:1 C) 2:3 D) 1:2 E) 3:1

9. The limiting reactant in a chemical reaction is the \_\_\_\_\_ that is \_\_\_\_\_.

- A) Reactant, in excess B) Product, left over  
 C) Reactant, used up D) Product, used up

10. Consider the following unbalanced reaction:  $\text{Cs} + \text{H}_2\text{O} \rightarrow \text{H}_2 + \text{CsOH}$

Which statement about the above reaction is true?

- A) 2 g of Cs will react with 2 moles of water

# Moles Test Chemistry Answer Key

**JL Elias**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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