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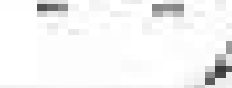

Molecular Geometry

Use the table to design for practice using the VSEPR model

Notes:

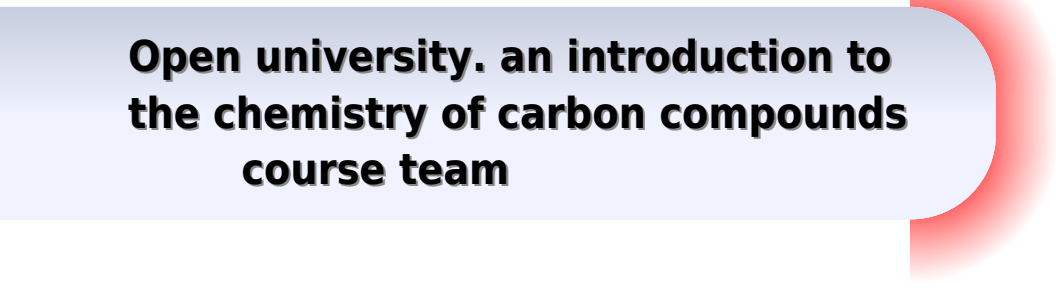
When you design your assignment, include an example of each, plus an additional example of the given molecule. Include Lewis structures, VSEPR model, and 3D representation. The 3D model is a molecular representation determined by physical and chemical properties for the molecule. Be different for each molecule from its previous representation.

Model 1 - Lewis Structures

Lewis Structures	Lewis Structure	3D Representation
CH_4 	CH_4 4 electron domains 0 bonding, 4 nonbonding	 Tetrahedral
H_2O 	H_2O 4 electron domains 2 bonding, 2 nonbonding	 Bent
CO_2 	CO_2 4 electron domains 4 bonding, 0 nonbonding	 Linear
NH_3 	NH_3 4 electron domains 3 bonding, 1 nonbonding	 Trigonal pyramidal
H_2S 	H_2S 4 electron domains 2 bonding, 2 nonbonding	 Bent
SO_2 	SO_2 4 electron domains 4 bonding, 0 nonbonding	 Bent
PCl_3 	PCl_3 4 electron domains 3 bonding, 1 nonbonding	 Trigonal pyramidal

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