

Mole Ratios

How can the coefficients in a chemical equation be interpreted?

NATALIA
LACAYO
P. 2

Why?

A balanced chemical equation can tell us the number of reactant and product particles (ions, atoms, molecules or formula units) that are necessary to conserve mass during a chemical reaction. Typically when we balance the chemical equation we think in terms of individual particles. However, in real life the reaction represented by an equation occurs an unimaginable number of times. Short of writing very large numbers (10^{23} or larger) in front of each chemical in the equation, how can we interpret chemical equations so that they more realistically represent what is happening in real life? In this activity you will explore the different ways a chemical reaction can be interpreted.

Model 1 – A Chemical Reaction



1. Consider the reaction in Model 1.

a. What are the coefficients for each of the following substances in the reaction?

1 N_2

3 H_2

2 NH_3

b. Draw particle models below to illustrate the reaction in Model 1.



2. Consider each situation below as it relates to the reaction in Model 1.

a. Calculate the amount of reactants consumed and products made.

b. Record the ratio of N_2 to H_2 to NH_3 . Reduce the ratio to the lowest whole numbers possible.

	N_2 Consumed	H_2 Consumed	NH_3 Produced	Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$ (reduced)
For a single reaction, how many molecules of each substance would be consumed or produced?	1	3	2	1:3:2
If the reaction occurred one hundred times, how many molecules would be consumed or produced?	100	300	200	1:3:2
If the reaction occurred 538 times, how many molecules would be consumed or produced?	538	1614	1076	1:3:2

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Sebastian Brünink

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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Chemistry James N. Spencer, George M. Bodner, Lyman H. Rickard, 2010-12-28 CHEMISTRY *SourceBook Version 2.1*, 1998 *Superhydrophobic Polymer Coatings* Sushanta Samal, Smita Mohanty, Sanjay Kumar Nayak, 2019-08-21

Superhydrophobic Polymer Coatings Fundamentals Design Fabrication and Applications offers a comprehensive overview of the preparation and applications of polymer coatings with superhydrophobicity guiding the reader through advanced techniques and scientific principles. Sections present detailed information on the fundamental theories and methods behind the preparation of superhydrophobic polymer coatings and demonstrate the current and potential applications of these materials covering a range of novel and marketable uses across industry including coatings with properties such as foul resistance and self cleaning anti icing and ice release corrosion inhibition antibacterial anti reflection slip and drag reduction oil water separation and advanced medical applications. This book is a highly valuable resource for academic researchers scientists and advanced students working on polymer coatings or polymer surface modifications as well as professionals across polymer science polymer chemistry plastics engineering and materials science. The detailed information in this book will also be of great interest to scientists R D professionals product designers and engineers who are looking to develop products with superhydrophobic coatings. Presents in depth information on the advanced methods required in the preparation of superhydrophobic polymer coatings. Covers the latest advances in the design of polymer coatings with superhydrophobic properties including nanofabrication. Explains cutting edge industrial and medical applications including self cleaning coatings corrosion inhibition anti icing and ice release and oil water separation. [Stable Isotopes in High Temperature Geological Processes](#) John W. Valley, Hugh P. Taylor, James R. O'Neil, 2018-12-17. Volume 16 of *Reviews in Mineralogy* introduces to high temperature stable isotope geochemistry and should provide an entry into the pertinent literature as well as some understanding of the basic concepts and potential applications. The first three chapters focus on the theory and experimental data base for equilibrium disequilibrium and kinetics of stable isotope exchange reactions among geologically important minerals and fluids. The fourth chapter discusses the primordial oxygen isotope variations in the solar system prior to formation of the Earth along with a discussion of isotopic anomalies in meteorites. The fifth chapter discusses isotopic variations in the Earth's mantle and the sixth chapter reviews the variations in the isotopic compositions of natural waters on our planet. In Chapters 7 8 9 and 10 these isotopic constraints and concepts are applied to various facets of the origin and evolution of igneous rocks bringing in much material on radiogenic isotopes as well because these problems require a multi dimensional attack for their solution. In Chapters 11 and 12 the problems of hydrothermal alteration by meteoric waters and ocean water are considered together with discussions of the physics and chemistry of hydrothermal systems and the ^{18}O ^{16}O history of ocean water. Finally in Chapters 13 and 14 these concepts are applied to problems of metamorphic petrology and ore deposits particularly with respect to the origins of the fluids involved in those processes.

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